

Write Titrating Volume (Spreads: 1.00)

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Acetic Acid (pKa = 4.75) titrated with NaOH

a. 0.100 M volume 25.00 mL acid base 0.100 M

b. 0.100 M volume 25.00 mL acid base 0.100 M

c. 0.100 M pH 4.75 compound that releases acetate
0.100 M acid base 0.100 M acid

Acetic Acid: CH_3COOH CH_3COO^- CH_3COO^-

CH₃COOH CH₃COO⁻ CH₃COO⁻
CH₃COOH CH₃COO⁻ CH₃COO⁻

Species	Acid (pH, Base pH, or Salt pH)	Acids with Mg (H^+ or H_2O)	Conductivity (C^+ or C^-)
CH_3COOH	H^+	H^+	C^+
CH_3COO^-	C^-	H^+	C^-
NaCl	H^+	H^+	C^-
NaOH	H^+	H^+	C^-

Note Taking Guide Episode 1101 Answers Key

RS Peters



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