

Write Titrating Volume (Spreads: 1.00)

Name _____

Acetic Acid (pKa = 4.75) Titration with NaOH

a. NaOH volume: 10.00 mL, pH: 11.00

b. NaOH volume: 5.00 mL, pH: 8.50

c. NaOH volume: 0.00 mL, pH: 2.50

Acetic Acid (pKa = 4.75) Titration with NaOH

NaOH volume: 10.00 mL, pH: 11.00
 NaOH volume: 5.00 mL, pH: 8.50
 NaOH volume: 0.00 mL, pH: 2.50

Species	Add (pH, Base pH, or Salt pH)	Acids with NaOH (pH = 10.0)	Conductivity (pH = 10.0)
HCl	1.0	1.0	1.0
HCl	1.0	1.0	1.0
HCl	1.0	1.0	1.0
HCl	1.0	1.0	1.0

Note Taking Guide 1101 Chemistry Answers

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