
The Periodic Law

Section Quiz: History of the Periodic Table

As Marjorie completed each question of this quiz, she placed that question's correct statement or best answer's letter in parentheses.

1. In developing his periodic table, Mendeleev's observations were
 - a. periodic elements.
 - b. elements with liquid states.
 - c. isotopes.
 - d. **properties.**
2. Mendeleev's periodic table showed that all elements in each of the columns shared many features, but grouped together elements with
 - a. **properties.**
 - b. similar structures.
 - c. masses.
 - d. isotopes.
3. Mendeleev predicted that the gaps in his periodic table represented
 - a. ions.
 - b. unknown elements.
 - c. unstable isotopes.
 - d. **undiscovered elements.**
4. The periodic table came before a periodic table based on increasing
 - a. atomic number.
 - b. **atomic mass.**
 - c. isotopes.
 - d. electronegativity.
5. The periodic law describes the physical and chemical properties of
 - a. **elements.**
 - b. isotopes.
 - c. ions.
 - d. compounds.

Periodic Law Section Quiz Answers

Karin Nielsen-Saines



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