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NOVA: Hunting the Elements

Part 1: Basic Chemistry

Periodic Table of Elements

Use the periodic table below to help you answer some questions covered in the documentary.

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| Na | Mg | 1 | | | | | | | | | | | Äl | Si | P | S | CI | Ar |
| ĸ | Ca | | Sc | Ti | V | Čr. | Mn | Fe | Co | Ni | Cu | Zn | Ga | Ge | As | Se | Br | Kr |
| Rb | Sr | | Ÿ | Žr | Nb | Mo | Tc | Ru | Rh | Pd | Ag | Cd | lin. | sn sn | Sb | Te | 53 | X€ |
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| Fr | Ra | 819-1042 96 96 | Lr | Rf | Db | Sg | Bh | Hs | Mit | 110 | Uuu | 110 | | Uuq | | | | |

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| La | Ce | Pr | Nd | Pm | Sm | Eu | Gd | Тb | Dy | Ho | Er | Tm | Ϋ́b |
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Gold - Au

* * Artistide, earlies

- Write the number of subatomic particles in gold:
 - a. Protons: 79
 - b. Neutrons: 118
 - e. Electrons: 79
- 2. Give one property of gold. shiny, malleable (soft), high density, electrically conductive
- 3. How much gold is in one ton of the mined rock?

Lounce

Nova Answer Hunting For Elements

Terry C. Jones

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