

Chapter 7: Chemical Reaction Guided Notes

I. Chemical Changes in Matter

A. Chemical Reaction

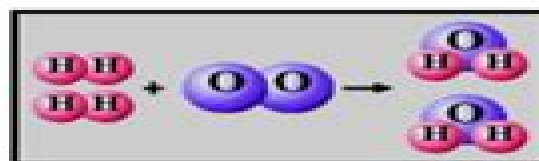
1. A change in which one or more substances are converted to _____ substances.



B. Law of Conservation of Matter

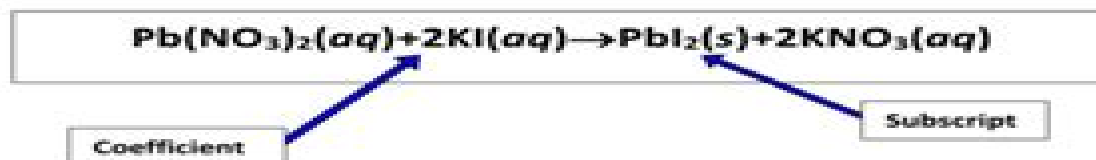
1. In a chemical reaction, matter is not _____ or _____.
2. Atoms can only be _____.
3. Discovered by Lavoisier.

_____ Hydrogen
 _____ Oxygen



_____ Hydrogen
 _____ Oxygen

C. Chemical Equations



1. _____ - # of units of each substance.
2. Individual atom = **atom**
 $2\text{Mg} \Rightarrow 2$ _____ of magnesium
3. Covalent substance = **molecule**
 $3\text{CO}_2 \Rightarrow 3$ _____ of carbon dioxide
4. Ionic substance = **unit**
 $4\text{MgO} \Rightarrow 4$ _____ of magnesium oxide

SYMBOL	MEANING
$\longrightarrow$	produces, forms
+	plus, and
(s)	solid
(l)	liquid
(g)	gas
(aq)	aqueous (solid dissolved in water)
Δ	the reactants are heated

Notetaking Guide Answer Keychemical Reactions

PT Brinkman



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