

Chemical Formula

Talking about proportions, molecules and relative of atoms indicated by a subscript.

Molecular $C_{10}H_{22}$ (molecules)

Formula represents the number of _____ of _____
in a single molecule.

Empirical C_5H_{11} (molecules)

Formula represents the proportions of _____ of
positive and negative _____, hence formula
unit.

Atomic Mass

The proportions that atoms combine through
mass together.

Masses

^{12}C mass 12g (atoms)

^{12}C mass proportionally charged
mass

Subscripts

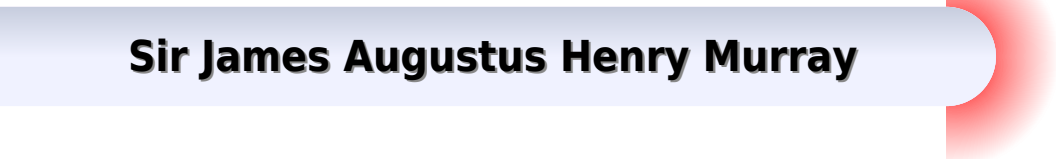
^{12}C (atoms)

^{12}C mass proportionally charged

^{12}C mass

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Sir James Augustus Henry Murray



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