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OWNERS MANUAL

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$$\text{Br}_2 + \text{C}_6\text{H}_{10}\text{O}_2 + 12\text{CO}_2 + 10\text{H}_2\text{O} \rightarrow 3\text{HBr} + 1\text{KHCO}_3 + 1\text{H}_2\text{O} + 1\text{KBr} + \text{CO}_2$$

$$\text{Br}_2 + 2\text{GaBr}_3 + 3\text{Na}_2\text{SO}_3 + 1\text{Ga}_2\text{SO}_3 + 3\text{NaBr} \rightarrow 5\text{SNO}_2 + 3\text{SnF}_2 + 1\text{N}_2\text{O}_3$$

 using the following equation

$$\text{NaOH} + \text{H}_2\text{SO}_4 \rightarrow 2\text{H}_2\text{O} + \text{Na}_2\text{SO}_4$$

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$$\text{HBr} + \text{KHCO}_3 \rightarrow \text{H}_2\text{O} + \text{KBr} + \text{CO}_2$$

$$\text{GaBr}_3 + \text{Na}_2\text{SO}_3 \rightarrow \text{Ga}_2\text{SO}_3 + 3\text{NaBr} + \text{SNO}_2 + 3\text{SnF}_2$$

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$$\text{SiO}_2 + 3\text{C} \rightarrow \text{SiC} + 2\text{CO}$$

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$$\text{HBr} + \text{KHCO}_3 \rightarrow \text{H}_2\text{O} + \text{KBr} + \text{CO}_2$$

$$\text{GaBr}_3 + \text{Na}_2\text{SO}_3 \rightarrow \text{Ga}_2\text{SO}_3 + 3\text{NaBr} + \text{SNO}_2 + 3\text{SnF}_2$$

$$\text{N}_2\text{O} + \text{CO}_2 \rightarrow \text{NaBr}$$

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