



Science Curriculum

Chemical Changes

Read the passage and answer the following questions:



1. What is the chemical formula for hydrogen peroxide?

Hydrogen peroxide is the chemical formula for the breaking of molecules. In this chemical reaction, the reactants are hydrogen peroxide (H_2O_2) and the products are water (H_2O) and oxygen (O_2). The reaction is exothermic, meaning it releases heat and energy.

2. What is the chemical formula for water? Is it a liquid, solid, or gas? Is it a reactant or a product? Is it a catalyst or an inhibitor?

3. What is the chemical formula for oxygen? Is it a reactant or a product?

Hydrogen peroxide is a chemical formula for the decomposition of hydrogen peroxide. The reactants are hydrogen peroxide (H_2O_2) and the products are water (H_2O) and oxygen (O_2). The reaction is exothermic, meaning it releases heat and energy. The reaction is also a redox reaction, meaning it involves the transfer of electrons.

4. According to the law of conservation of mass, if the mass of the reactants is 100 grams, what is the mass of the products? Is it 100 grams, 200 grams, or 50 grams?

100 grams



5. What is the chemical formula for water? Is it a liquid, solid, or gas?

6. What is the chemical formula for oxygen? Is it a reactant or a product?

7. What is the chemical formula for hydrogen peroxide? Is it a reactant or a product?

Hydrogen peroxide is a chemical formula for the decomposition of hydrogen peroxide. The reactants are hydrogen peroxide (H_2O_2) and the products are water (H_2O) and oxygen (O_2).

8. What is the chemical formula for water? Is it a liquid, solid, or gas?

9. What is the chemical formula for oxygen? Is it a reactant or a product?

Hydrogen peroxide is a chemical formula for the decomposition of hydrogen peroxide. The reactants are hydrogen peroxide (H_2O_2) and the products are water (H_2O) and oxygen (O_2).

10. According to the law of conservation of mass, if the mass of the reactants is 100 grams, what is the mass of the products? Is it 100 grams, 200 grams, or 50 grams?

100 grams

Reinforcement Chemical Changes

Shasha Hu



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