

1. Circle the element with the largest atomic radius and put a square around the element with the smallest atomic radius:

Cu **K** Ni **Br**

- a. Explain why you made these choices: *All of the elements are in the same period. The trend in atomic radius as you go across a period is DECREASING. Therefore, the element on the far left (K) is the largest, and the element on the far right (Br) is the smallest.*

2. Circle the element with the highest ionization energy and put a square around the element with the lowest ionization energy:

Cu **K** Ni **Br**

- a. Explain why you made these choices: *All of the elements are in the same period. The trend in ionization energy as you go across a period is INCREASING. Therefore, the element on the far left (K) has the lowest ionization energy, and the element on the far right (Br) has the highest ionization energy.*

3. Circle the element with the highest electronegativity and put a square around the element with the lowest electronegativity:

Cu **K** Ni **Br**

- a. Explain why you made these choices: *All of the elements are in the same period. The trend in electronegativity as you go across a period is INCREASING. Therefore, the element on the far left (K) has the lowest electronegativity, and the element on the far right (Br) has the highest electronegativity.*

4. For each of the following groups: Circle the element with the largest atomic radius and put a square around the element with the smallest atomic radius:

- 5.
- O C **Be** **Ne** Same Period
 - Na Rb **Fr** **H** Same Group
 - Pb** **C** Sn Si Same Group
 - Au W S **Fr** **Ne** Zn Challenge

6. For each of the following groups: Circle the element with the highest ionization energy and put a square around the element with the lowest ionization energy:

- O C **Be** **Ne** Same Period
- Na Rb **Fr** **H** Same Group
- Pb** **C** Sn Si Same Group
- Au W S **Fr** **Ne** Zn Challenge

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SJ Ball



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small the amount which may be present as an ingredient or an ingredient of a compound ingredient or the substance is used as a food additive or an ingredient or component of such a substance or a substance or food us

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web august 24 2022 any severe reaction to food that elicits an immunological response is defined as a food allergy in today s world food allergies are very common with their prevalence spreading to many countries around the globe affecting most children 4 6 compared to adults 1 2

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web the use of precautionary allergen labelling there will be occasions during the food manufacturing process where it is not possible to avoid the unintended presence of allergens to address this precautionary allergen labelling pal was introduced to communicate the risk of unintended presence of allergens in foods to consumers

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web allergen declarations refers to required information on food packaging where certain allergens are present e g in the ingredient list and or a contains statement while pal and advisory labelling refers to statements that indicate the possible presence of allergens as a consequence of allergen cross

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web allergen advice statements can also be used on the product label to explain how allergen information is presented on a label for example allergen advice for allergens see ingredients

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web dec 31 2020 the list above contains the correct naming of all 14 allergens as well as the allergen s letter code as recommended by the codex commission it is possible to mark dishes e g on a restaurant menu with a codex letter code or a

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