Structure of an atom

 List the three subatomic particles which make up an atom. Include their location, charge, and atomic mass.

a.	Electron	Negative charge	Electron Cloud
be.	Proton	Positive charge	Nucleus
C.	Neutron	Neutral charge	Nucleus

- In a neutral atom, which two subatomic particles are equal in number to each other?
 Protons and Electrons
- 3. What does the atomic number of an element indicate?

The number of protons in the atom of that element

4. What is an isotope?

Atoms with the same number of protons, but different masses due to a different number of neutrons

5. Will two different atoms of carbon always have the same number of protons?

Element key

13 Al 26.982 Aluminum

- 1. What are the number of protons, neutrons, and electrons in an atom of aluminum?
 - a. Protons = 13
 - b. Neutrons = 14
 - c. Electrons = 13

Bohr Model

- 1. How do you know how many energy levels (orbits) a Bohr model will have?
 - a. By the period the element is located in
- How many electrons can each energy level hold?

 $1^{10} - 2$

2 - 8

 $3^{nd} - 18$

The Periodic Table

- 1. How is the periodic table arranged? Increasing atomic number
- 2. What are the vertical columns 2 called? Groups or Families
- What are the horizontal rows ← called? Periods
- 4. Why are elements put into groups? Same number of valence electrons
- 5. What is a valence electron? Number of electrons on the outer energy level (shell)
- 6. Know how many valence electrons each group has
- 7. Which group of metals are highly reactive and why? Alkali Metals. Only 1 ver to lose
- 8. Which group of non-metals are highly reactive and why? Halogens. Only 1 we to gain
- Name some physical properties of metals: luster, ductile, malleable, conductor of heat and electricity
- 10. Name some physical properties of non metals: no luster, brittle, not a good conductor
- What are metalloids and where are they found on the periodic table? Share properties of metals and metalloids. On zig-zag line starting at Boron.

Periodic Table Study Guide Answers

Dr. Sunita Bhagiya, ,Er. Meera Goyal

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Table of Contents Periodic Table Study Guide Answers

- 1. Understanding the eBook Periodic Table Study Guide Answers
 - The Rise of Digital Reading Periodic Table Study Guide Answers
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Periodic Table Study Guide Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Table Study Guide Answers
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Periodic Table Study Guide Answers
 - Personalized Recommendations
 - Periodic Table Study Guide Answers User Reviews and Ratings
 - Periodic Table Study Guide Answers and Bestseller Lists
- 5. Accessing Periodic Table Study Guide Answers Free and Paid eBooks

- Periodic Table Study Guide Answers Public Domain eBooks
- Periodic Table Study Guide Answers eBook Subscription Services
- o Periodic Table Study Guide Answers Budget-Friendly Options
- 6. Navigating Periodic Table Study Guide Answers eBook Formats
 - o ePub, PDF, MOBI, and More
 - Periodic Table Study Guide Answers Compatibility with Devices
 - Periodic Table Study Guide Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodic Table Study Guide Answers
 - Highlighting and Note-Taking Periodic Table Study Guide Answers
 - Interactive Elements Periodic Table Study Guide Answers
- 8. Staying Engaged with Periodic Table Study Guide Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Table Study Guide Answers
- 9. Balancing eBooks and Physical Books Periodic Table Study Guide Answers
 - Benefits of a Digital Library
 - o Creating a Diverse Reading Collection Periodic Table Study Guide Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Periodic Table Study Guide Answers
 - Setting Reading Goals Periodic Table Study Guide Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Periodic Table Study Guide Answers
 - Fact-Checking eBook Content of Periodic Table Study Guide Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
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