

CHAPTER 5 REVIEW*The Periodic Law***SECTION 1****SHORT ANSWER** Answer the following questions in the space provided.

1. c In the modern periodic table, elements are ordered
 - (a) according to decreasing atomic mass.
 - (b) according to Mendeleev's original design.
 - (c) according to increasing atomic number.
 - (d) based on when they were discovered.
2. d Mendeleev noticed that certain similarities in the chemical properties of elements appeared at regular intervals when the elements were arranged in order of increasing
 - (a) density.
 - (b) reactivity.
 - (c) atomic number.
 - (d) atomic mass.
3. b The modern periodic law states that
 - (a) no two electrons with the same spin can be found in the same place in an atom.
 - (b) the physical and chemical properties of an element are functions of its atomic number.
 - (c) electrons exhibit properties of both particles and waves.
 - (d) the chemical properties of elements can be grouped according to periodicity, but physical properties cannot.
4. c The discovery of the noble gases changed Mendeleev's periodic table by adding a new
 - (a) period.
 - (b) series.
 - (c) group.
 - (d) level.
5. d The most distinctive property of the noble gases is that they are
 - (a) metallic.
 - (b) radioactive.
 - (c) metalloid.
 - (d) largely unreactive.
6. c Lithium, the first element in Group 1, has an atomic number of 3. The second element in this group has an atomic number of
 - (a) 4.
 - (b) 10.
 - (c) 11.
 - (d) 18.
7. An isotope of fluorine has a mass number of 19 and an atomic number of 9.
 - 9 a. How many protons are in this atom?
 - 10 b. How many neutrons are in this atom?
 - ${}^{19}_{9}\text{F}$ c. What is the nuclear symbol of this fluorine atom, including its mass number and atomic number?

Periodic Law Section Quiz Answers

Ying-Ying Zheng

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

Periodic Law Section Quiz Answers:

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biological science are examined including Mendeleev's Periodic Law Kekule's structure for benzene the light quantum hypothesis quantum mechanics chromosome theory and natural selection In general it is found that theories are accepted for a combination of successful predictions and better explanations of old facts Making 20th Century Science is a large scale historical look at the implementation of the scientific method and how scientific theories come to be accepted **Score-well ICSE Model Test Papers Chemistry Class 10** Amar Nath Bhutani, 2025-08-27 The Score Well ICSE Model Test Papers for Class 10 Chemistry offers a complete exam preparation package through carefully designed specimen style test papers Covering all units Periodic Table Bonding Electrolysis Acids Bases Salts Organic Chemistry Metallurgy and Numerical Problems the papers help students master both theory and application Each set comes with detailed solutions examiner guidelines and step by step numerical solving methods enabling students to avoid errors and strengthen accuracy Regular practice of these model tests develops speed confidence and problem solving skills making this book a must have for securing top grades in ICSE Chemistry

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