

2006 Pontiac Torrent Owner Manual

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Owners Manual Pontiac

Puja Mehta



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$$\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$$

$$\text{Mg}(\text{OH})_2 + 2\text{HNO}_3 \rightarrow \text{Mg}(\text{NO}_3)_2 + 2\text{H}_2\text{O}$$

$$2\text{NaClO}_3 \rightarrow 2\text{NaCl} + 3\text{O}_2$$

$$4\text{Al} + 3\text{O}_2 \rightarrow 2\text{Al}_2\text{O}_3$$

$$\text{N}_2 + \text{O}_2 \rightarrow 2\text{NO}$$

$$2\text{NH}_3 + 3\text{O}_2 \rightarrow 2\text{N}_2 + 6\text{H}_2\text{O}$$

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$$\text{Mg}(\text{OH})_2 + \text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2\text{O}$$
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 metal acid salt hydrogen

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$$2\text{C}_2\text{H}_2 + 5\text{O}_2 \rightarrow 4\text{CO}_2 + 2\text{H}_2\text{O}$$
 if 1.0 mol of C_2H_2 is burned how many moles of CO_2 are formed
 if 1.0 mol of C_2H_2 is burned how many moles of oxygen gas are consumed

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