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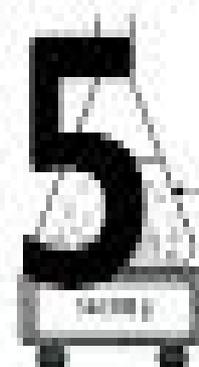
MEMO

marks, $\text{mgCO}_2/\text{gH}_2\text{O}$, and $\text{mgCO}_2/\text{gH}_2\text{O}$ due to the effect of concentration on the rate.

$\text{mgCO}_2 = 15000 - 0,002t$

The apparatus used for this investigation is shown below.

2025



Flask with reaction mixture

Science

- 1.1 Define the term rate of reaction.
- 1.2 Write down ONE controlled variable for this investigation.

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JR Anderson



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