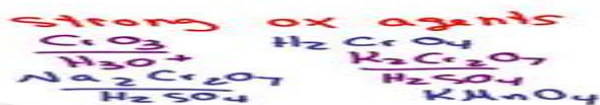


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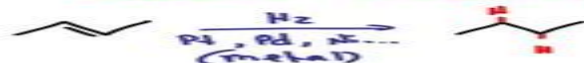
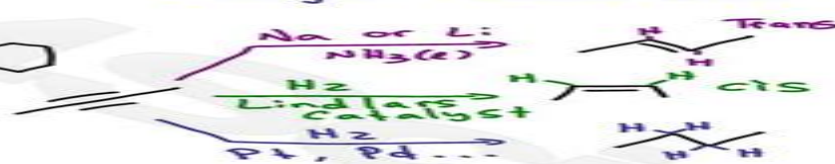
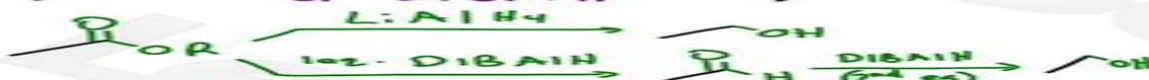
CC=CC $\xrightarrow[\text{KOH, } \text{O}_4]{\text{OsO}_4}$ CC(O)CC(O)C



Chemical reaction showing the oxidation of 4-ethyl-2-methylstyrene using KMnO_4 to produce 4-carboxy-2-methylstyrene.

only 1° & 2°
get oxidized
3° = N/R

Alkene Reduction

c1ccccc1>[Li].[Na]>c1ccccc1CC(=O)O>>CCO
$$Li^+ + H - \overset{H}{\underset{H}{\underset{H}{|}{|}{|}{A}}}-H$$
$$\text{NaBH}_4 = \text{weak}$$


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