

# BOARD QUESTION PAPER: MARCH 2022

## CHEMISTRY

Time: 3 Hrs

Max. Marks: 70

### General Instructions:

The question paper is divided into four sections.

- (1) **Section A:** Q. No. 1 contains Ten multiple choice type of questions carrying One mark each.  
Q. No. 2 contains Eight very short answer type of questions carrying One mark each.
- (2) **Section B:** Q. No. 3 to Q. No. 14 are Twelve short answer type of questions carrying Two marks each. (Attempt any Eight).
- (3) **Section C:** Q. No. 15 to Q. No. 26 are Twelve short answer type of questions carrying Three marks each. (Attempt any Eight).
- (4) **Section D:** Q. No. 27 to Q. No. 31 are Five long answer type of questions carrying Four marks each. (Attempt any Three).
- (5) Use of log table is allowed. Use of calculator is not allowed.
- (6) Figures to the right indicate full marks.
- (7) For each multiple choice type of question, it is mandatory to write the correct answer along with its alphabet, e.g. (a)...../(b)...../(c)...../(d)..... etc.

No marks(s) shall be given, if ONLY the correct answer or the alphabet of the correct answer is written. Only the first attempt will be considered for evaluation.

### SECTION – A

Q.1. Select and write the correct answer for the following multiple choice type of questions:

[10]

- (i) The co-ordination number of atoms in body centred cubic structure (bcc) is \_\_\_\_\_.  
(a) 4                      (b) 6                      (c) 8                      (d) 12
- (ii) In calculating osmotic pressure, the concentration of solute is expressed in \_\_\_\_\_.  
(a) molarity                      (b) molality  
(c) mole fraction                      (d) percentage mass
- (iii) The enthalpy change for the chemical reaction  $\text{H}_2\text{O}_{(s)} \longrightarrow \text{H}_2\text{O}_{(g)}$  is called enthalpy of \_\_\_\_\_.  
(a) vapourisation                      (b) fusion  
(c) combustion                      (d) sublimation
- (iv) Which of the following transition element shows maximum oxidation state?  
(a) Sc                      (b) Fe                      (c) Mn                      (d) V
- (v) The correct formula for the complex compound, sodium hexacyanoferrate (III) is \_\_\_\_\_.  
(a)  $\text{Na} [\text{Fe}(\text{CN})_6]$                       (b)  $\text{Na}_2 [\text{Fe}(\text{CN})_6]$   
(c)  $\text{Na}_3 [\text{Fe}(\text{CN})_6]$                       (d)  $\text{Na}_4 [\text{Fe}(\text{CN})_6]$

# Oct 2014chemistry Q Paper Maharashtra

**Padma V. Devarajan, Prajakta  
Dandekar, Anisha A. D'Souza**



**Oct 2014chemistry Q Paper Maharashtra:**

This book delves into Oct 2014chemistry Q Paper Maharashtra. Oct 2014chemistry Q Paper Maharashtra is a crucial topic that must be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Oct 2014chemistry Q Paper Maharashtra, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:

- Chapter 1: Introduction to Oct 2014chemistry Q Paper Maharashtra
- Chapter 2: Essential Elements of Oct 2014chemistry Q Paper Maharashtra
- Chapter 3: Oct 2014chemistry Q Paper Maharashtra in Everyday Life
- Chapter 4: Oct 2014chemistry Q Paper Maharashtra in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, the author will provide an overview of Oct 2014chemistry Q Paper Maharashtra. The first chapter will explore what Oct 2014chemistry Q Paper Maharashtra is, why Oct 2014chemistry Q Paper Maharashtra is vital, and how to effectively learn about Oct 2014chemistry Q Paper Maharashtra.

3. In chapter 2, the author will delve into the foundational concepts of Oct 2014chemistry Q Paper Maharashtra. The second chapter will elucidate the essential principles that need to be understood to grasp Oct 2014chemistry Q Paper Maharashtra in its entirety.

4. In chapter 3, the author will examine the practical applications of Oct 2014chemistry Q Paper Maharashtra in daily life. This chapter will showcase real-world examples of how Oct 2014chemistry Q Paper Maharashtra can be effectively utilized in everyday scenarios.

5. In chapter 4, this book will scrutinize the relevance of Oct 2014chemistry Q Paper Maharashtra in specific contexts. The fourth chapter will explore how Oct 2014chemistry Q Paper Maharashtra is applied in specialized fields, such as education, business, and technology.

6. In chapter 5, this book will draw a conclusion about Oct 2014chemistry Q Paper Maharashtra. This chapter will summarize the key points that have been discussed throughout the book.

The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Oct 2014chemistry Q Paper Maharashtra.

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