#### Fluid and Electrolytes: Potter and Perry Chapter 42

- Overview
  - Fluid, electrolyte, and acid-base balances within the body maintain health and function of all body systems
  - <u>Fluid</u>: water that contains dissolved or suspended substances such as glucose, mineral salts, and proteins
    - Characteristics of body fluids influence body system function and have regulatory mechanisms that keep them in balance:
      - Fluid amount à volume
      - Concentration à osmolality
      - Composition à electrolyte concentration
      - Degree of acidity à pH
  - Water = 60% of adult body weight
    - Decreases with age
  - Fluid distribution: movement of fluid among its various compartments
    - Fluid located in 2 compartments:
      - Extracellular (ECF): outside of the cell
        - Intravascular liquid part of the blood (plasma)
        - Interstitial located between the cells and outside the blood vessels (within the tissues)
      - Intracellular (ICF): inside of the cell
    - Body tries to maintain homeostasis
  - Osmolality: measure of the # of particles per kilogram of water
    - Normal range: 285 295 mOsm/kg H2O
    - Tonicity: concentration of solutes in solution
      - Isotonic same tonicity as normal blood (no fluid shift)
      - Hypotonic more dilute than the blood
      - Hypertonic more concentration than normal blood
  - Electrolytes: a compound that separates into ions when it dissolves in water
    - Cations: + charged
      - Na, K, Ca, Mg
    - Anions: negatively charged
      - Cl. HCO3
- Movement of Water and Electrolytes:
  - Active transport; requires energy (ATP) to move electrolytes across cell
    membranes against concentration gradient (areas of low concentration to areas
    of high concentration)
    - Example: Sodium potassium pump
      - Moves Na out of a cell and K into it, keeping ICF lower in NA and higher in K than the ECF

# **Potter And Perry Chapter 42**

**Gabby Koutoukidis, Jodie Hughson** 

## **Potter And Perry Chapter 42:**

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