

More Atomic Structure Practice: Use the periodic table to help you find the answers!

Draw atomic structures and electron dot diagrams for the following atoms: H, He, Li, Be, B, C, O, F, Na, Ne, Al, Si, P, Cl, Ar

Here are the atomic structures: NOTE! Only the electrons in their shells are shown below! The number of protons equals the TOTAL number of electrons. The number of neutrons is found below.

1st Shell	H 1 						He 2 	
2nd Shell	Li 3 	Be 4 	B 5 	C 6 	N 7 	O 8 	F 9 	Ne 10 
3rd Shell	Na 11 	Mg 12 	Al 13 	Si 14 	P 15 	S 16 	Cl 17 	Ar 18 

Here are the electron dot diagrams:

I	II							III	IV	V	VI	VII	O
H •													He ••
Li •	• Be •							• B •	• C •	• N •	• O •	• F •	• Ne ••
Na •	• Mg •							• Al •	• Si •	• P •	• S •	• Cl •	• Ar ••
K •	• Ca •							• Ga •	• Ge •	• As •	• Se •	• Br •	• Kr ••
Rb •	• Sr •							• In •	• Sn •	• Sb •	• Te •	• I •	• Xe ••
Cs •	• Ba •							• Tl •	• Pb •	• Bi •	• Po •	• At •	• Rn ••

Metal
Metalloid
Nonmetal

How many neutrons are in the following atoms (show your work!): H, He, Li, Be, B, C, O, F, Ne, Na, Al, Si, P, Cl, Ar

Atomic mass (rounded) — Atomic Number = # neutrons

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