

Voyager PRO UC

Revolutionizing the mobile office. Be truly mobile with the intelligent, next generation Voyager PRO UC. Unify PC and mobile phone communications while experiencing Smart Sensor technology to help you stay connected — and productive — wherever work takes you.

NEXT GENERATION TECHNOLOGY

New Smart Sensor technology for a seamless and intuitive user experience

- Automatically answers calls simply by placing on your ear
- Automatically transfers calls between mobile phone and headset
- When on a mobile or PC call, softphone presence is automatically updated
- Eliminates accidental dialing by locking call button when headset is not worn

OUTSTANDING AUDIO QUALITY

- Dual-mic AudioIQ² DSP for noise cancellation, delivering clear voice quality to you and your listener
- Advanced WindSmart[®] technology provides three layers of wind protection
- PC Wideband delivers heightened speech clarity, providing the best possible PC audio quality
- Integrated A2DP lets you listen to streaming media, including your favorite songs, podcasts, turn-by-turn navigation and more.

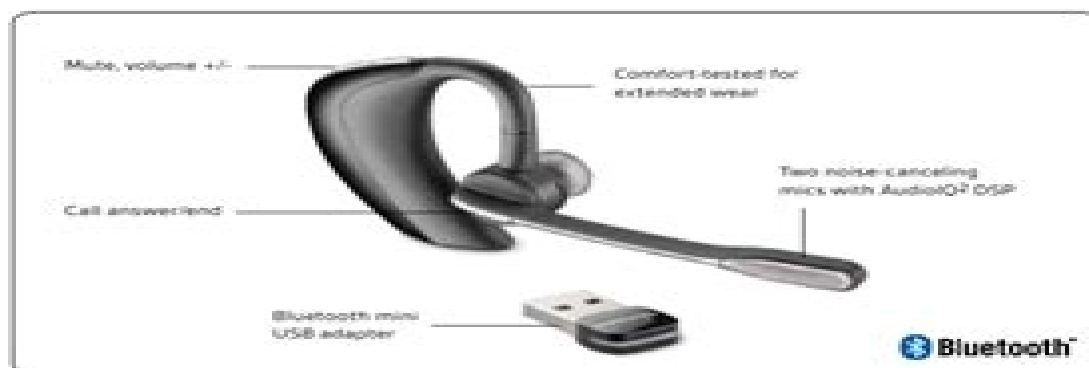
SUPERIOR CALL MANAGEMENT

- Multi-device connectivity lets you manage PC and mobile phone calls from a single headset
- New Bluetooth mini USB adapter can stay in your laptop port for always-ready connectivity
- Enhanced voice alerts announce remaining talk time, connection status, battery level and mute

GLOBAL CUSTOMER CARE

Backed by Plantronics, industry leading global service and support.

800-544-4660 (US and Canada)
831-458-7700 (outside US)



OPTIMIZED FOR UNIFIED COMMUNICATIONS

Connects to	PC and mobile phone
Ideal for	Users in office and on the road with a mix of PC and mobile phone communication for conference calls, webinars, video conferencing and listening to multimedia
Talk time/standby	Up to 6 hrs/5 days
Digital encryption	128-bit
Headset weight	17.5 grams
Wireless frequency	Bluetooth v2.1 + EDR
Audio performance	Wideband: up to 6,800 Hz; echo cancellation; A2DP
Hearing protection	SoundGuard [®] : Sound leveling for listening comfort and acoustic limiting for protection against sounds above 118 dBA
Limited warranty	1 year
Models	B230: UC Standard version built for UC applications and softphones from Avaya, Cisco, IBM and more B230-M: Version optimized for Microsoft [®] Lync [™] and Microsoft OCS 2007

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Raffaela Di Napoli



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this video accompanies the chm 252 lab nitration of bromobenzene in this video studens will get an introduction to the lab as well as the set up for conducting the reaction students should

analysis of nitration of bromobenzene 779 words cram - Jul 18 2022

for the nitration 4 01 ml of nitric acid 4 00 ml of sulfuric acid and 4 56 ml bromobenzene were used to complete the reaction with these measurements bromobenzene was found to be the limiting reagent since only 4 56 ml of bromobenzene was used the amount of product predicted to be recovered was 8 7533 g

lab 334 nitration of aromatic compounds preparation of studocu - Dec 23 2022

the aim of the experiment is to determine and study electrophilic aromatic substitution reaction eas particularly nitration of methyl benzoate go give methyl m nitrobenzoate thus determine the weight melting point and percentage yield of the pure methyl m nitrobenzoate abstract

experiment synthesis of 1 bromo 4 nitrobenzene pbworks - May 28 2023

in this experiment we will carry out the nitration of bromobenzene bromine substituted on an aromatic ring is an ortho para directing group it will direct an incoming electrophile to the ortho and para positions it is also deactivating meaning that its reactions will be slower than the identical reactions of benzene

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section 20711 lab experiment nitration of bromobenzene introduction the chemical reaction of nitration consists of a nitro group being added to or substituted in a molecule nitration can basically be carried out by a mixture of concentrated nitric acid and sulfuric acid this mixture is useful to obtain the active nitronium ion

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1 go to the restroom and wash the affected area with lots of soap and water the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area 3 the summary to this warning is not to touch any part of your body while performing this experiment if you must leave the lab for any reason 1

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in the nitration of bromobenzene a mixture of sulfuric acid and nitric acid react with one another nitric acid the weaker of the two acids will serve as base to form nitronium ion bromobenzene will attack the nitronium ion by nucleophilic attack

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nitration of bromobenzene lab report improved essays 731 words 3 pages open document essay sample check writing quality show more this experiment was performed in order to demonstrate the principles of nitration through electrophilic aromatic substitution it also examined the effects of bromine on an aromatic ring ortho para vs meso directing

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then actual is your actual findings yield actual theoretical $\times 100$ what are the possible side reactions of nitration of bromobenzene dinitration can occur at ortho para or at both ortho positions occurs in overheating meta product happens very rarely because bromine is an ortho para director

[nitration of bromobenzene gravity waves](#) - Nov 09 2021

the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area theory and explanation of the nitration of bromobenzene if the temperature of the reaction mixture exceeds 60 °C

experiment 3 electrophilic aromatic substitution reactions studocu - Jan 24 2023

the bromine substituent in bromobenzene is an ortho para director the para position is favored due to steric effects in the reaction that takes place in this lab a nitro group is substituted onto bromobenzene at the para position to form 1-bromo-4-nitrobenzene

nitration lab flashcards quizlet - Sep 19 2022

study with quizlet and memorize flashcards containing terms like reacting sulfuric and nitric acid together produces the nitronium ion alone nitric acid is not a strong enough electrophile to react with bromobenzene sulfuric acid is added to the nitric acid the reverse rxn can overheat and boil easily includes a claisen adaptor and

nitration of bromobenzene 6 nitration of bromobenzene 10 19 - Oct 01 2023

over 5 minutes slowly pipet this bromobenzene into the icy cold nitration mixture HNO_3 3 h 2 so 4 in the 125 ml flask and stir continue to stir the reaction mixture in the ice bath for 30 minutes do not heat in a new 250 ml erlenmeyer

[experiment 11 electrophilic aromatic substitution nitration](#) - Jan 12 2022

on the right the arenium cation has a nitro set with incredibly small 0.6 angstrom bond distances this will result in an optimization failure as the atoms are even close for the programme to accommodate their electrons demonstration of creating the p arenium cation by the nitration of bromobenzene

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lab report 15 nitration of bromobenzene brynn abbott amp katie muerer bea0006 uah ch 336 02 september 21 2022 abstract in this experiment the electrophilic aromatic substitution mechanism was used to obtain 4-nitrobromobenzene and 2-nitrobromobenzene recrystallization was used to purify 4-nitrobromobenzene

[nitration of bromobenzene the purpose of this lab was to](#) - Jul 30 2023

the purpose of this lab was to observe where substituents are added on a benzene ring by way of electrophilic aromatic

substitution the electrophilic source in the nitration of bromobenzene comes from a nitronium ion NO_2^+ which acts as the electrophilic species

nitration of bromobenzene lab report docx lab report 10 - Oct 21 2022

nitration of bromobenzene is an electrophilic aromatic substitution reaction the electrophile will be formed from 4 0ml of nitric acid and 4 0ml of sulfuric acid and will then react with 4 5ml of bromobenzene this will produce o bromonitrobenzene and p bromonitrobenzene

experiment 11 electrophilic aromatic substitution nitration - Feb 10 2022

experimental spectra reaction a fluorobenzene stock fluorobenzene 1 h 1 h 19 f 19 f 13 c nmr fid for reference only sample 1 h 1 h 19 f 19 f 13 c nmr eas nitration of fluorobenzene crude product not available for submission for credit

nitration of bromobenzene chem 154 gwu studocu - Mar 26 2023

lab writeup the purpose of this lab was to observe where substituents are added on benzene ring way of electrophilic aromatic substitution the electrophilic

5 nitration lab docx lab 5 nitration of bromobenzene - Aug 19 2022

chem 2261 lab 5 nitration of bromobenzene introduction an electrophilic aromatic substitution reaction is an important process that allows for intermediate chemical structures to be used as precursors to pharmaceuticals and other industrially produced chemicals by adding different functional groups into an aromatic ring

lab report nitration of bromobenzene ipl org - Feb 22 2023

lab report 10 nitration of bromobenzene raekwon filmore cm 244 section 40 march 27 2018 introduction for this experiment nitration of bromobenzene was the focus of the lab the benzene is an aromatic compound and when it reacts with wither a mixture of sulfuric acid or nitric acid creates what is known as a nitro group