

Step 8. Periodic table trends

Electronegativity

Ionization energy

He

Atomic radius

The periodic table displays various elements with their symbols and atomic numbers. Trends are indicated by colored arrows:

- Electronegativity:** Indicated by a black arrow pointing right across the top.
- Ionization energy:** Indicated by a green arrow pointing right across the top and a green arrow pointing up on the right side.
- Atomic radius:** Indicated by a red arrow pointing down on the left side and a red arrow pointing left across the bottom.
- Non-metallic properties:** Indicated by a purple arrow pointing up and to the right.
- Metallic properties:** Indicated by a blue arrow pointing down and to the right.

Ionization Energy

Electronegativity

Atomic radius

An arrow head shows the INCREASE of the property

Example: comparing K atom with Br atom

- Atomic radius of K is HIGHER than atomic radius of Br
- Ionization energy of K is LOWER than ionisation energy of Br
- Metallic properties of K are HIGHER than metallic properties of Br



Periodic Trends Review Answer Key

Sabine Zange



Periodic Trends Review Answer Key:

Decoding **Periodic Trends Review Answer Key**: Revealing the Captivating Potential of Verbal Expression

In a period characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its capability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Periodic Trends Review Answer Key**," a mesmerizing literary creation penned by a celebrated wordsmith, readers attempt an enlightening odyssey, unraveling the intricate significance of language and its enduring effect on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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