

Periodic Trends Worksheet

Use the periodic table and your knowledge of periodic trends to answer the following questions.

1. Which atom in each pair has the larger atomic radius?
a) Li or **K** b) **Ca** or Na c) **Ga** or B d) O or **C** e) Cl or **Br**
f) Be or **Ba** g) **Si** or S h) Fe or **Au**

2. What is the periodic trend for atomic size from top to bottom in a group? from left to right in a period?

Top to bottom: increases due to adding energy levels Left to Right: decreases

3. Why do atoms get smaller as you move left to right in a period?
due to increased nuclear charge and no added shielding

4. Which element in each pair has a larger ionization energy?
a) Na or **O** b) **Be** or Ba c) **Ar** or F d) **Cu** or Ra e) I or **Ne**
f) K or **V** g) **Ca** or Fr h) W or **Se**

5. Explain the relationship between the relative size of an ion to its neutral atom and the charge on the ions.
Positive ions are smaller than the neutral atom it comes from, negative ions are larger than the neutral atom it comes from

6. Which particle has the larger radius in each atom/ion pair?
a) Na, Na⁺ b) S, S²⁻ c) I, I⁻ d) **Al**, Al³⁺

7. What is ionization energy? What is first ionization energy?

The amount of energy needed to remove an electron from an atom. First ionization energy is the energy required to remove a single valence electron from an atom.

8. What is the periodic trend for first ionization energy?

It increases going from left to right due to increased nuclear charge, and decreases going down a group due to increased distance from the nucleus and increased shielding.

9. Arrange the following groups of elements in order of increasing ionization energy.
a) Be, Mg, Sr b) Bi, Cs, Ba c) Na, Al, S
high, middle, low high, low, middle low, middle, high

10. Which element in each pair has a higher electronegativity value?
a) Cl, **F** b) C, **N** c) **Mg**, Ne d) As, Ca

11. What is the periodic trend for electronegativity? Increases going left to right, decreases going down.

Periodic Trends Pre Lab Answers

L Towner

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the L Towner text.

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Exploring General Chemistry in the Laboratory Colleen F. Craig, Kim N. Gunnerson, 2017-02-01 This laboratory manual is intended for a two semester general chemistry course The procedures are written with the goal of simplifying a complicated and often challenging subject for students by applying concepts to everyday life This lab manual covers topics such as composition of compounds reactivity stoichiometry limiting reactants gas laws calorimetry periodic trends molecular structure spectroscopy kinetics equilibria thermodynamics electrochemistry intermolecular forces solutions and coordination complexes By the end of this course you should have a solid understanding of the basic concepts of chemistry which will give you confidence as you embark on your career in science *Chemistry* McGraw-Hill Staff, 2001-07 Prentice Hall

Chemistry, 2000 **Integrated Approach to Coordination Chemistry** Rosemary A. Marusak, Kate Doan, Scott D. Cummings, 2007-03-07 Coordination chemistry is the study of compounds formed between metal ions and other neutral or negatively charged molecules This book offers a series of investigative inorganic laboratories approached through systematic coordination chemistry It not only highlights the key fundamental components of the coordination chemistry field it also exemplifies the historical development of concepts in the field In order to graduate as a chemistry major that fills the requirements of the American Chemical Society a student needs to take a laboratory course in inorganic chemistry Most professors who teach inorganic chemistry laboratory prefer to emphasize coordination chemistry rather than attempting to cover all aspects of inorganic chemistry because it keeps the students focused on a cohesive part of inorganic chemistry which has applications in medicine the environment molecular biology organic synthesis and inorganic materials

Sustainable Green Chemistry Mark Anthony Benvenuto, 2017-04-10 Sustainable Green Chemistry the 1st volume of Green Chemical Processing covers several key aspects of modern green processing The scope of this volume goes beyond bio and organic chemistry highlighting the ecological and economic benefits of enhanced sustainability in such diverse fields as petrochemistry metal production and wastewater treatment The authors discuss recent progresses and challenges in the implementation of green chemical processes as well as their transfer from academia to industry and teaching at all levels Selected successes in the greening of established processes and reactions are presented including the use of switchable polarity solvents actinide recovery using ionic liquids and the removal of the ubiquitous bisphenol A molecule from effluent streams by phytodegradation

Guided Inquiry Experiments for General Chemistry Nancy K. Kerner, Ram S. Lamba, 2007-10-19 The use of the laboratory is a valuable tool in developing a deeper understanding of key chemical concepts from the experimental process This lab manual encourages scientific thinking enabling readers to conduct investigations in chemistry It shows how to think about the processes they are investigating rather than simply performing a laboratory experiment to the specifications set by the manual Each experiment begins with a problem scenario and ends with questions requiring feedback on the problem **Laboratory Manual for Principles of General Chemistry** Jo Allan

Beran,2010-11-01 This new edition of the Beran lab manual emphasizes chemical principles as well as techniques The manual helps students understand the timing and situations for the various techniques The Beran lab manual has long been a market leading lab manual for general chemistry Each experiment is presented with concise objectives a comprehensive list of techniques and detailed lab intros and step by step procedures **Science Spectrum** Holt Rinehart & Winston,Holt, Rinehart and Winston Staff,2003-03 The World of Chemistry Jones,Jones Staff,1996 **Exploring Physical Science in the Laboratory** John T. Salinas ,2019-02-01 This full color manual is designed to satisfy the content needs of either a one or two semester introduction to physical science course populated by nonmajors It provides students with the opportunity to explore and make sense of the world around them to develop their skills and knowledge and to learn to think like scientists The material is written in an accessible way providing clearly written procedures a wide variety of exercises from which instructors can choose and real world examples that keep the content engaging Exploring Physical Science in the Laboratory guides students through the mysteries of the observable world and helps them develop a clear understanding of challenging concepts *Principles Of Descriptive Inorganic Chemistry* Gary Wulfsberg,1991-05-29 This is the only text currently available organized by class of compound and by property or reaction type not group by group or element by element which requires students to memorize isolated facts This is the only text currently available organized by class of compound and by property or reaction type not group by group or element by element which requires students to memorize isolated facts Translated into Italian **Laboratory Manual** Jo A. Beran,1990 **ERDA Energy Research Abstracts** United States. Energy Research and Development Administration,1976 **Linne & Ringsrud's Clinical Laboratory Science - E-Book** Mary Louise Turgeon,2015-02-10 Using a discipline by discipline approach Linne perforated pages make it easy for students to remove for use in the lab and for assignment of review questions as homework NEW Instrumentation updates show new technology being used in the lab NEW Additional key terms in each chapter cover need to know terminology NEW Additional tables and figures in each chapter clarify clinical lab science concepts *Manual of Molecular and Clinical Laboratory Immunology* John L. Schmitz,Barbara Detrick,Maurice R. G. O'Gorman,2024-11-15 THE authoritative guide for clinical laboratory immunology For nearly 50 years the Manual of Molecular and Clinical Laboratory Immunology has been the premier resource for laboratories students and professionals involved in the clinical and technical details of diagnostic immunology testing The 9th Edition continues its tradition of providing comprehensive clinical and technical information on the latest technologies used in medical and diagnostic immunology Led by a world renowned group of authors and editors this new edition reflects substantial changes aimed at improving and updating the Manual s utility while reflecting the significant transformations that have occurred since the last edition including the revolution of gene editing and the widespread adoption of molecularly engineered cellular therapies Topical highlights include Laboratory Management three new chapters cover essential aspects of quality assurance quality improvement and quality management aligning with the

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LaValle,2013-07-02 OVER 47 000 COPIES SOLD If you're like most people you probably rely on your doctor to interpret the results of your blood tests which contain a wealth of information on the state of your health A blood test can tell you how well your kidneys and liver are functioning your potential for heart disease and diabetes the strength of your immune system the chemical profile of your blood and many other important facts about the state of your health And yet most of us cannot decipher these results ourselves nor can we even formulate the right questions to ask about them that is until now In *Your Blood Never Lies* best selling author Dr James LaValle clears the mystery surrounding blood test results In simple language he explains all the information found on a typical lab report the medical terminology the numbers and percentages and the laboratory jargon and makes it accessible This means that you will be able to look at your own blood test results and understand the significance of each biological marker being measured To help you take charge of your health Dr LaValle also recommends the most effective standard and complementary treatments for dealing with any problematic findings Rounding out the book are explanations of lab values that do not appear on the typical blood test but that should be requested for a more complete picture of your current physiological condition A blood test can reveal so much about your body but only if you can interpret the results *Your Blood Never Lies* provides the up to date information you need to take control of your health *The Shock and Vibration Digest* ,1986 **Applied Mechanics Reviews** ,1982 *Selected Water Resources Abstracts* ,1989

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