# CRUDE OIL

Teck 1 - Crude oil key word definition	and the

Crude oil is a finite resource found in rocks. It is the remains of start & unimal life that was buried in mud many millions of years ago.

a) Define finite resource ......

Crude oil is a mixture of a very large number of compounds. Most of The compounds in crude all are hydrocorbons De Fine Hydrocentees

#### Tesk Z - Hydrocarbons

a) Highlight which of the following one but hydrocorbons. Explain your added ....

#### Test 3 - Alteres

Alkones are the simplest type of hydrocorbon. They contain only single Tobbs of albanes.

Nome	Melecular formula	Displayed formula
methone	CH4	H-Ç-H
_	C/HL	
propone	C)H4	
Sylane		

Viscosity of hydrocerton  Flammability  c) What property of the molecules does frestional distillation rely upon to separate them from crude oil?  () Siteman is collected at the bottom of the column. It is still a liquid & oild not turn into a vapour when heared. What does that fall you about the boiling point of bitamen compared to the rest of the hydrocertons in crude oil?  (2) Choose the correct word from the word bank to complete this description of the process of frectional distillation separates the crude oil into different	THERE &	Fractional distillation:			ter pro-	Anthony person	Botton
bo) Add labels to the diagram apposite to make clear what happens as you go up the column to the following:  - Size of melecula:  - Boiling point of hydrocarbon  - Viscosity of hydrocarbon  - Viscosity of hydrocarbon  - Planmability  () What property of the meleculas does frectional distritorion  - ridy open to separate them from crude oil?  () Situmen is collected at the bottom of the column. It is still  - right of aid not turn into a uppor when heated. What does  - that tail you about the boiling point of bitumen compared to  - the rest of the hydrocarbons in anyole oil?  - a) Choose the correct word from the word bonk to complete this description of the process of  - frectional distribution.  - Frectional distribution separates the crude oil into different models.  - Size each frection contains a mixture of different-sized hydrocarbons  - Size each frection contains a mixture of different-sized hydrocarbons with similar boiling points.  - The larger malecules will have greater.  - The wardones these forces. This means these chain hydrocarbons will bell of  - The larger malecules will have greater.  - The wardones these forces.  - This means these	c) What	is the process of fractional distillet	fon used for?		also.	[ent]	
c) Add labels to the diagram appears to make alsor what heppens as you go so the column to the following:  Size of malessia:  Soliding point of hydrocarbon  Flammability  () What property of the malessias does frectioned distribution maly upon to separate them from crude oil?  (i) Bitamen is collected at the bottom of the column. It is still a liquid & did not turn into a vipour when heated. What does that fall you about the boiling point of alternat componed to the nest of the hydrocarbon in crude oil?  (ii) Choose the correct word from the word bonk to complete this description of the process of frectional distribution separates the crude oil into different coording to the of the hydrocarbon in crude oil into different coording to the of the different hydrocarbona.  So each fraction contains a mixture of different-sized hydrocarbons with similar boiling points.  The larger malessies will have greater to evantome these feroes. This means these these hydrocarbons will bell at temperatures.  Differentiable forces helper fraces higher fractions have belief belling points along the sample larger.  Task 5 - Conception:  Hydrocarbons are used as faels as they release a lat of energy when they are bornt.  White the word equation to show the combustion of a hydrocarbon fuel:  HYDROCARBON +			CONTRACTOR OF THE PARTY OF THE	100	T.	The Party of the P	100
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