

Salters OCR B chemistry EL not mine

2 bond pairs, 0 lone pairs - ANSLinear

180°

e.g. BeCl₂

3 bond pairs, 0 lone pairs - ANSTrigonal Planar

120°

e.g. BF₃

2 bond pairs, 1 lone pair - ANSBent

118°

e.g. NO₂

4 bond pairs, 0 lone pairs - ANSTetrahedral

109.5°

e.g. CH₄

3 bond pairs, 1 lone pair - ANSPyramid

107°

e.g. NH₃

2 bond pairs, 2 lone pairs - ANSBent

104.5°

e.g. H₂O

5 bond pairs, 0 lone pairs - ANSTrigonal Bipyramid

120°, 90°

e.g. PCl₅

6 bond pairs, 0 lone pairs - ANSOctahedral

90°

e.g. SF₆

In 1803 John Dalton said... - ANSThat all atoms are SPHERES and all elements are made from different spheres

Plum pudding model - ANSIn 1897 J.J. Thompson discovered the electron and said there were negative electrons "plums" in a positive "pudding"

Gold Leaf Experiment - ANS1909, Ernest Rutherford.

Fired positive alpha particles at gold leaf. Most went through to other side, some deflected back.

He concluded that the atom had a very small positively charged nucleus and a negative cloud of empty space.

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Frank Harriss



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Table of Contents Ocr Salters B Chemistry Skill 3 All Specimens

1. Understanding the eBook Ocr Salters B Chemistry Skill 3 All Specimens
 - The Rise of Digital Reading Ocr Salters B Chemistry Skill 3 All Specimens
 - Advantages of eBooks Over Traditional Books
2. Identifying Ocr Salters B Chemistry Skill 3 All Specimens
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ocr Salters B Chemistry Skill 3 All Specimens
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ocr Salters B Chemistry Skill 3 All Specimens
 - Personalized Recommendations
 - Ocr Salters B Chemistry Skill 3 All Specimens User Reviews and Ratings
 - Ocr Salters B Chemistry Skill 3 All Specimens and Bestseller Lists
5. Accessing Ocr Salters B Chemistry Skill 3 All Specimens Free and Paid eBooks

- Ocr Salters B Chemistry Skill 3 All Specimens Public Domain eBooks
 - Ocr Salters B Chemistry Skill 3 All Specimens eBook Subscription Services
 - Ocr Salters B Chemistry Skill 3 All Specimens Budget-Friendly Options
6. Navigating Ocr Salters B Chemistry Skill 3 All Specimens eBook Formats
- ePub, PDF, MOBI, and More
 - Ocr Salters B Chemistry Skill 3 All Specimens Compatibility with Devices
 - Ocr Salters B Chemistry Skill 3 All Specimens Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Ocr Salters B Chemistry Skill 3 All Specimens
 - Highlighting and Note-Taking Ocr Salters B Chemistry Skill 3 All Specimens
 - Interactive Elements Ocr Salters B Chemistry Skill 3 All Specimens
8. Staying Engaged with Ocr Salters B Chemistry Skill 3 All Specimens
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ocr Salters B Chemistry Skill 3 All Specimens
9. Balancing eBooks and Physical Books Ocr Salters B Chemistry Skill 3 All Specimens
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ocr Salters B Chemistry Skill 3 All Specimens
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Ocr Salters B Chemistry Skill 3 All Specimens
- Setting Reading Goals Ocr Salters B Chemistry Skill 3 All Specimens
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Ocr Salters B Chemistry Skill 3 All Specimens
- Fact-Checking eBook Content of Ocr Salters B Chemistry Skill 3 All Specimens
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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