# **Graphing Periodic Trends Worksheet**

 Using the ionization energy (IE) values of the elements in Period 2 listed below, make a line graph of IE vs. the atomic number.

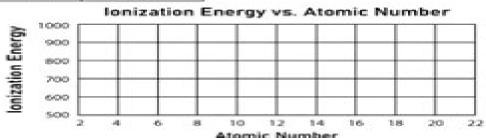
			Period 2
Element	Atomic Number	IE (kJ/mol)	
Li	3	519	
Вe	4	900	
В	5	799	
C	6	1088	

Element	Atomic Number	IE (kJ/mol)
Li	3	519
Be	4	900
В	5	799
C	6	1088

2500								
2000 -	_	_	_	-	+-	+-	+-	+
1500	_	-+-	_		-	+-	+	-
1000	_	-	-		+-	-	+-	
500	_		-		-	-	-	-
L								

- a) What trend do you notice for the ionization energies in Period 2?
- b) Explain why you observe this trend.
- Make a scatter plot of ionization energy vs. atomic number for the first three atoms in Group 2A listed below.

Group 2A				
Element	Atomic Number	IE (kJ/mol)		
Be	4	900		
Mg	12	736		
Ca	20	590		



# **Reading Periodic Trend Graphing Questions**

**D** Kirk

#### **Reading Periodic Trend Graphing Questions:**

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