

## 10.4 Electrophilic Reactions

**General Introduction** An electrophile is an electron-seeking species, reacting with a lone pair or a double bond.

### Electrophilic Substitution

Electrophilic Substitution reactions are reactions in which an electrophile reacts with a nucleophile.

- It forms an intermediate called a carbocation (one of the two common types of carbocation is the  $\text{sp}^2$  carbocation).
- The carbocation is then attacked by a nucleophile, which forms a new bond, called a carbocation.

Electrophilic Substitution reactions are also known as electrophilic substitution reactions.

- The carbocation intermediate is then attacked by a nucleophile, which forms a new bond, called a carbocation.
- The carbocation is then attacked by a nucleophile, which forms a new bond, called a carbocation.

When reacting a nucleophile with an electrophile, the following reactions occur:

### Electrophilic Substitution

- In the following reaction, the nucleophile attacks the carbocation intermediate, forming a carbocation intermediate, which then reacts with a nucleophile, forming a carbocation intermediate.
- When the carbocation intermediate reacts with a nucleophile, a carbocation intermediate is formed.
- When the carbocation intermediate reacts with a nucleophile, a carbocation intermediate is formed.
- When the carbocation intermediate reacts with a nucleophile, a carbocation intermediate is formed.

- When the carbocation intermediate reacts with a nucleophile, a carbocation intermediate is formed.
- When the carbocation intermediate reacts with a nucleophile, a carbocation intermediate is formed.

### Transition

- When the carbocation intermediate reacts with a nucleophile, a carbocation intermediate is formed.

# Reading And Study Workbook 19 4 Neutralization Reactions

**L Cohen**



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