

Topic	Grade	What do you need to know	 or 	What do You have to work on?
Atomic Structure	A	- Why an atom is neutral in terms of sub-atomic particles - Outline isotopes from data about numbers of protons and neutrons		
	B	- Know the charges and masses of protons, electrons and neutrons. - Know how isotopes are the same but have different mass number.		
	C	- Describe the arrangement of an atoms electron structure - 2.8.8.2 - Describe the arrangement of elements in the periodic table.		
Ionic Bonding	A	- Explain ionic bonding using the 'dot and cross' model. - Explain the melting point of NaCl by the structure of and bonding. - Explain why the melting point of NaCl is lower than that of MgO.		
	B	- Predict the properties of substances with giant ionic structures. - Understand that atoms with a full outer shell are electronically stable.		
	C	- Ionic bonds are formed from metals and non-metals - Electrons are donated and received. - Metals always donate the electron(s). - Non- metals always receive the electron(s)		
The Periodic Table	A	- Explain some covalent bonds using a 'dot and cross' model - Explain double covalent binds using a 'dot and cross' model		
	B	Know why CO₂ and H₂O have low melting points		
	C	- Covalent bonds always have non-metals and share electrons - Know that the group number is the same as the number of electrons in the outer shell. - Be able to find out how many shells an atom has from the period it is in.		
The Group 1 Elements	A	- Construct a balanced symbol equation for Cu²⁺, Fe²⁺ and Fe³⁺ to make Cu(OH)₂, Fe(OH)₂, Fe(OH)₃. - Reactivity with water gives NaOH and KOH. - Know the process is oxidation.		
	B	- Why group 1 elements form similar positive ions. - Construct a balance symbol equation of group 1 with water		
	C	- Metal + Water $\longrightarrow$ Metal Hydroxide + Hydrogen - Know how to use a flame test to identify potassium, lithium and sodium compounds.		
The Group 7 Elements	A	- Predict if displacement reactions will occur with other halogens - Explain the trend of reactivity of group 7 by ideas of electron gain - Explain why a process is reduced from its ionic		

Ocr Gateway Separate Sciences B C4 Checklist

Marco Cascella



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familiarise students with the types of questions that they may come across in the exam Feel confident with thorough exam preparation for GCSE OCR Gateway Science A Including different question formats this exam practice workbook allows you to revise in a way that suits you best and improve your performance in the exam Included in this book a variety of question formats as used by OCR to give students the best possible practice prior to tests and exams lots of quick tests to check understanding relevant formulae and data sheets clear and concise coverage of the exam assessed content simple and engaging explanations highlighted key words plus a supporting glossary to build vocabulary **Essentials - OCR Gateway GCSE Additional Science** Natalie King,2011-09-01 This guide covers all the exam assessed content on the OCR Gateway additional science specification It is designed to help students get the most out of their revision Information is broken down into manageable chunks and higher tier material is clearly differentiated on the page **OCR Gateway Additional Science** Natalie King,2013-01-01 OCR Gateway GCSE Additional Science Essentials covers all the exam assessed content on the OCR Gateway Additional Science specification It is designed to help students get the most out of their revision information is broken down into manageable chunks and higher tier material is clearly differentiated on the page *OCR Gateway Science* Natalie King,2013-01-01 OCR Gateway GCSE Science Essentials covers all the exam assessed content on the OCR Gateway Science specification It is designed to help students get the most out of their revision information is broken down into manageable chunks and higher tier material is clearly differentiated on the page **Wright Science Pocket Booster - OCR Gateway a Biology 4, 5 And 6** Vicki Wright,2019-03-07 The Wright Science Pocket Booster is designed to assist those students embarking on their GCSE Science examinations This book has been written to cover the key facts for the first biology exam paper for the higher tier on OCR Gateway A Inside the book in addition to the printed notes you also have QR codes that will take you directly to a video that supports that topic This book is written for the separate science GCSE Biology course It has been written by a secondary science teacher teaching in England using the resources my students have always found to be helpful Essentials - OCR Gateway GCSE Additional Science Natalie King,Sam Holyman,Claire Hutchinson,2011 *Gcse Pp Science Ocr Gateway Higher* ,2007-07-01 GCSE Practice Papers provide focused exam preparation and practice The GCSE OCR Gateway Science Science Higher pack contains three papers which follow the format of the actual tests enabling pupils to become familiar with their content and style *OCR Gateway GCSE Biology* ,2016

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