

Polyatomic Ion Worksheet

Place the charges on each ion – Write the Formula – Name the Compound

Remember – The cation always goes first.

Example:	NH_4^{+1}	&	I^{-1}	Formula:	<u>NH_4I</u>	Name:	<u>Ammonium Iodide</u>
1.	NH_4	&	O	Formula:	_____	Name:	_____
2.	Ca	&	CO_3	Formula:	_____	Name:	_____
3.	HCO_3	&	Na	Formula:	_____	Name:	_____
4.	NO_3	&	Mg	Formula:	_____	Name:	_____
5.	K	&	SO_4	Formula:	_____	Name:	_____
6.	Li	&	PO_4	Formula:	_____	Name:	_____

Finding the Formula from the Name

Example:	Lithium Chloride	Ions:	<u>Li^{+1} & Cl^{-1}</u>	Formula:	<u>LiCl</u>
1.	Sodium Oxide	Ions:	_____	Formula:	_____
2.	Potassium Bromide	Ions:	_____	Formula:	_____
3.	Calcium Nitride	Ions:	_____	Formula:	_____
4.	Beryllium Sulfide	Ions:	_____	Formula:	_____
5.	Magnesium Iodide	Ions:	_____	Formula:	_____
6.	Aluminum Oxide	Ions:	_____	Formula:	_____
7.	Ammonium Bromide	Ions:	_____	Formula:	_____
8.	Calcium Bicarbonate	Ions:	_____	Formula:	_____
9.	Potassium Phosphate	Ions:	_____	Formula:	_____
10.	Aluminum Sulfate	Ions:	_____	Formula:	_____

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Tao Wei



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Table of Contents Polyatomic Ions Worksheet With Answers Physical Science If8767

1. Understanding the eBook Polyatomic Ions Worksheet With Answers Physical Science If8767
 - The Rise of Digital Reading Polyatomic Ions Worksheet With Answers Physical Science If8767
 - Advantages of eBooks Over Traditional Books
2. Identifying Polyatomic Ions Worksheet With Answers Physical Science If8767
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Polyatomic Ions Worksheet With Answers Physical Science If8767
 - User-Friendly Interface
4. Exploring eBook Recommendations from Polyatomic Ions Worksheet With Answers Physical Science If8767
 - Personalized Recommendations
 - Polyatomic Ions Worksheet With Answers Physical Science If8767 User Reviews and Ratings
 - Polyatomic Ions Worksheet With Answers Physical Science If8767 and Bestseller Lists
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 - Polyatomic Ions Worksheet With Answers Physical Science If8767 Budget-Friendly Options

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 - ePub, PDF, MOBI, and More
 - Polyatomic Ions Worksheet With Answers Physical Science If8767 Compatibility with Devices
 - Polyatomic Ions Worksheet With Answers Physical Science If8767 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Polyatomic Ions Worksheet With Answers Physical Science If8767
 - Highlighting and Note-Taking Polyatomic Ions Worksheet With Answers Physical Science If8767
 - Interactive Elements Polyatomic Ions Worksheet With Answers Physical Science If8767
8. Staying Engaged with Polyatomic Ions Worksheet With Answers Physical Science If8767
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Polyatomic Ions Worksheet With Answers Physical Science If8767
9. Balancing eBooks and Physical Books Polyatomic Ions Worksheet With Answers Physical Science If8767
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Polyatomic Ions Worksheet With Answers Physical Science If8767
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Polyatomic Ions Worksheet With Answers Physical Science If8767
 - Setting Reading Goals Polyatomic Ions Worksheet With Answers Physical Science If8767
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Polyatomic Ions Worksheet With Answers Physical Science If8767
 - Fact-Checking eBook Content of Polyatomic Ions Worksheet With Answers Physical Science If8767
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

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web 3 developing a responsible offer and involving consumers in implementing the food transition carrefour wants to inform engage and involve customers in order to make them stakeholders in the food transition carrefour highlights the food transition in stores through the deployment of its global communications programme act for food

