## Naming Compounds Containing Polyatomic Ions

Use the following list of polyatomic ions to name the compounds given.

	OH NO <sub>3</sub> C <sub>2</sub> H <sub>3</sub> O <sub>3</sub>	hydroxide nitrate acetate	CIO <sub>3</sub> HCO <sub>3</sub> SO <sub>4</sub> CO <sub>3</sub>	chlorate bicarbonate sulfate carbonate	SO <sub>3</sub> PO <sub>4</sub> NH <sub>4</sub>	sulfite phosphate ammonium
1.	CuSO			26. NaOH		
22.	KCIO,					
3.	NH4OH					
4.	K2CO3					
5	Na <sub>3</sub> SO <sub>4</sub>					
6	. KC <sub>2</sub> H <sub>2</sub> O <sub>2</sub>			31. KHCO,		
7.	. NH,Cl			32. NH,F		
8.	Mg(OH) <sub>2</sub>			33. KCiO <sub>3</sub>		
9.	Zn(NO <sub>3</sub> ) <sub>2</sub>			34. NaC <sub>2</sub> H <sub>3</sub> O <sub>2</sub>		
	кон			37. Be(OH) <sub>2</sub>		
				39. Al <sub>2</sub> (SO <sub>4</sub> ) <sub>5</sub> _		
				41. NH,I		
	MgSO <sub>4</sub>			42. NH, HCO,		
	AgNO <sub>3</sub>			43. Cu <sub>2</sub> (PO <sub>4</sub> ) <sub>2</sub>		
	Na <sub>2</sub> CO <sub>5</sub>			44. Ni(OH)		
20.	NaHCO <sub>s</sub>					
	Ca(NO <sub>3</sub> ) <sub>2</sub>			46. MnCO <sub>3</sub>		
24.	LiNO,					
25.	Al(NO <sub>3</sub> ) <sub>3</sub>					

# **Polyatomic Ions Naming Answer**

**R Pring** 

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