

Simplifying Radicals with Variables

Date _____ Period _____

Simplify.

1) $\sqrt{16x^2}$

2) $\sqrt{32b^4}$

3) $\sqrt{512a^2}$

4) $\sqrt{48x^2}$

5) $\sqrt{20v}$

6) $\sqrt{72p^2}$

7) $\sqrt{196n^2}$

8) $\sqrt{196p^3}$

9) $\sqrt{18n^3}$

10) $\sqrt{216n^2}$

11) $\sqrt{80p^2}$

12) $\sqrt{75a^4}$

13) $\sqrt{48b^3}$

14) $\sqrt{18x^2y}$

15) $\sqrt{18xy}$

16) $\sqrt{8m^2n^2}$

17) $\sqrt{36a^2b}$

18) $\sqrt{27xy^2}$

19) $\sqrt{108x^4y^2z^3}$

20) $\sqrt{18p^4q^2r^2}$

21) $\sqrt{12mn^2p^2}$

22) $\sqrt{32h^2j^2k}$

23) $\sqrt{32abc}$

24) $\sqrt{64x^2yz}$

25) $\sqrt{32x^2yz}$

Prentice Hall Algebra 1 Chapter10 Simplifying Radicals

JL Elias



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buffers typically consist of an acid base pair with the acid and base differing by the presence or absence of a proton a conjugate acid base pair for instance one of the buffers that maintain the ph of human blood involves carbonic acid H_2CO_3 and its conjugate base the bicarbonate ion HCO_3^-

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an acid and a base react not to produce a salt and a solvent but to form a new acid and a new base the concept of neutralization is thus absent brønsted lowry acid base behavior is formally independent of any solvent making it more all encompassing than the arrhenius model

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acid base titration difference between acid and base acids and bases definition an acid is any hydrogen containing substance that is capable of donating a proton hydrogen ion to another substance a base is a molecule or ion able to accept a hydrogen ion from an acid acidic substances are usually identified by their sour taste

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