

Name _____ Date _____ Class _____

17

THERMOCHEMISTRY

SECTION 17.1 THE FLOW OF ENERGY—HEAT AND WORK (pages 505–510)

This section explains the relationship between energy and heat, and distinguishes between heat capacity and specific heat.

► Energy Transformations (page 505)

- What area of study in chemistry is concerned with the heat transfers that occur during chemical reactions? thermochemistry
- Where the use of energy is concerned (in a scientific sense), when is work done?
Work is done when a force is used to move an object.
- Circle the letter next to each sentence that is true about energy:
 - ☒ Energy is the capacity for doing work or supplying heat.
 - ☒ Energy is detected only because of its effects.
 - ☐ Heat is energy that transfers from one object to another because they are at the same temperature.
 - ☒ Gasoline contains a significant amount of chemical potential energy.
- Circle the letter next to each sentence that is true about heat.
 - ☒ One effect of adding heat to a substance is an increase in the temperature of that substance.
 - ☐ Heat always flows from a cooler object to a warmer object.
 - ☒ If two objects remain in contact, heat will flow from the warmer object to the cooler object until the temperature of both objects is the same.

► Exothermic and Endothermic Processes (pages 506–507)

- What can be considered the “system” and what are the “surroundings” when studying a mixture of chemicals undergoing a reaction? Write your answers where indicated below.

System: The mixture of chemicals itself is considered the system.

Surroundings: Everything but the mixture of chemicals is the surroundings, but practically speaking, the immediate vicinity of the system.

Prentice Hall Chemistry Chapter 11 Answers

Roman Wölfel



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Equilibrium and Solutions James E. Banks, 1967 **Study Guide and Solutions Manual for Organic Chemistry** Susan McMurry, John McMurry, 1988 **Study Guide and Solutions Manual to Accompany Fundamentals of Organic Chemistry** John McMurry, Susan McMurry, 1986 *Clinical Chemistry - E-Book* Donna Larson, 2015-12-17

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High Pressure Technology Spain,1977-11-01

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Quantum Chemistry Walter Kauzmann, 2013-10-22 Quantum Chemistry An Introduction provides information pertinent to the fundamental aspects of quantum mechanics This book presents the theory of partial differentiation equations by using the classical theory of vibrations as a means of developing physical insight into this essential branch of mathematics Organized into five parts encompassing 16 chapters this book begins with an overview of how quantum mechanical deductions are made This text then describes the achievements and limitations of the application of quantum mechanics to chemical problems Other chapters provide a brief survey of some essential properties of the associated Legendre functions The final chapter deals with the Franck Condon principle which states that transitions tend to occur between vibrational levels of two different electronic states for which either the minimum or maximum values of the internuclear distance in the potential energy diagram occur with the same nuclear configuration This book is a valuable resource for chemists

Volume Properties Emmerich Wilhelm, Trevor Letcher, 2014-12-17 Volumetric properties play an important role in research at the interface of physical chemistry and chemical engineering but keeping up with the latest developments in the field demands a broad view of the literature Presenting a collection of concise focused chapters this book offers a comprehensive guide to the latest developments in the field and a starting point for more detailed research The chapters are written by acknowledged experts covering theory experimental methods techniques and results on all types of liquids and vapours The editors work at the forefront of

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