Answers for Predicting Products of Chemical Reactions

For all combustion reactions of hydrocarbons or hydrogen-carbon-oxygen molecules, the products will always be CO: and H:O (assuming a complete reaction).

a. Combustion:	C_6H_{12}	+	901	-	6CO ₃	*	6H;O
b. Combustion:	2C ₄ H ₆	+	1102	\rightarrow	8CO2	+	6H:O
c. Combustion:	$C_0H_{10}O_3$	94	707		6CO ₂		5H ₂ O

1. Synthesis: Mg + I₂ → MgI:

Note that Mg can only form Mg2 and I can only form I', so the product will must have a 1:2 cation anion ratio.

2. Double displacement: CuCl₂ + H₂S → CuS + 2HCl Note that the product is **not** H₂Cl₂. It is important to recognize that CuCl₂ is made of three ions, Cu¹⁺ and two Cl².

Double displacement: NaOH + HCIO₄ → NaCIO₄ + H₂O
 In this question, you must recognize that perchlorate, ClO₄, and hydroxide, OH¹, are polyatomic ions and will not break apart. Also, this is an acid-base reaction, so the products should be salt and water.

Decomposition: ZnCO₂ + heat → ZnO + CO₂.
 When reactions have heat as a reactant, it is very likely that they will involve decompositions. Carbonate compounds usually decompose to CO₂ and a metal oxide.

5. Single replacement: 2HCl + Zn → ZnCl₂ + H₂
Note that one reactant is in its elemental form; if a single replacement reaction is going to occur, the species NOT in its elemental form in the reactants (H¹ in this case), will end up in its elemental form in the products (H₂) and the species that IS in its elemental form (Zn) will end up ionized (Zn²). Note that zinc can only form a Zn² ion, so it will have two chloride ions. Note also that hydrogen in its elemental form is H₂, not H.

6. <u>Single replacement</u> 2Na + MgCl₂ → 2NaCl + Mg Again notice that one species is in its elemental form (Na). The magnesium in MgCl₂ is an ion (Mg²⁺), but is transformed into its elemental state (Mg), while the Na is converted into an ion (Na⁺, sodium only forms a +1 ion).

Double replacement CaCl: + K:CO → CaCO: + 2KCl
 Recognize that carbonate is a polyatomic ion (CO₁²) and that the cations are already stable ions (Ca²² and K²).

8. <u>Synthesis</u> 2K + CI: → 2KCI

Note that both materials are elemental species, so the only result could be a synthesis reaction. In the product, K^{*} and CF are formed. Note that we would NOT write the product as K₂Cl₂. Ionic compounds are written in their simplest forms.

Double replacement 3BaCl: + 2K:PO₁ → Ba:(PO₁): + 6KCl
 Note that phosphate (PO₁) is a polyatomic ion and will not break apart. Since barium is a +2 ion, the barium phosphate will have a 3:2 ratio of Ba:PO₁ in order to balance the charge.

10. <u>Double replacement</u> H:SO_a + 2KOH → 2H:O + K:SO_a Notice that sulfate (SO_a²) is a polyatomic ion and that potassium exists as a +1 ion (K²). Also, this is an acid-base reaction, so the products should be salt and water.

11. Decomposition Al₂(CO₂): + beat → Al₂O₂ + 3CO₂ Another decomposition that will generate CO₂ and a metal oxide (note that Al forms a +3 ion, Al²⁻, and monatomic oxygen will have a -2 charge, O²).

Predicting Products Answer Key

Fun Man Fung, Christoph Zimmermann

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Organic Chemistry David R. Klein, 2020-12-22 In Organic Chemistry 4th Edition Dr David Klein builds on the phenomenal success of the first three editions with his skills based approach to learning organic chemistry The Klein program covers all the concepts typically covered in an organic chemistry course while placing a special emphasis on the skills development needed to support these concepts Students in organic chemistry need to be able to bridge the gap between theory concepts and practice problem solving skills Klein's SkillBuilder examples and activities offer extensive opportunities for students to develop proficiency in the key skills necessary to succeed in organic chemistry Carson-Dellosa Publishing, 2015-03-16 Chemistry for grades 9 to 12 is designed to aid in the review and practice of chemistry topics Chemistry covers topics such as metrics and measurements matter atomic structure bonds compounds chemical equations molarity and acids and bases The book includes realistic diagrams and engaging activities to support practice in all areas of chemistry The 100 Series science books span grades 5 to 12 The activities in each book reinforce essential science skill practice in the areas of life science physical science and earth science. The books include engaging grade appropriate activities and clear thumbnail answer keys Each book has 128 pages and 100 pages or more of reproducible content to help students review and reinforce essential skills in individual science topics. The series will be aligned to current science Chemistry, 2015-03-16 Chemistry for grades 9 to 12 is designed to aid in the review and practice of chemistry standards topics Chemistry covers topics such as metrics and measurements matter atomic structure bonds compounds chemical equations molarity and acids and bases The book includes realistic diagrams and engaging activities to support practice in all areas of chemistry The 100 Series science books span grades 5 to 12 The activities in each book reinforce essential science skill practice in the areas of life science physical science and earth science. The books include engaging grade appropriate activities and clear thumbnail answer keys Each book has 128 pages and 100 pages or more of reproducible content to help students review and reinforce essential skills in individual science topics. The series will be aligned to current science IBM Solution Developer Practice Tests: 350 Practice Questions & Detailed Explanations for Exam Success standards CloudRoar Consulting Services, 2025-08-15 The IBM Solution Developer Practice Tests 350 Practice Questions Detailed Explanations for Exam Success is your essential guide to mastering the IBM Solution Developer certification exam This certification is a testament to your expertise in developing testing and integrating applications using IBM s robust suite of tools and platforms Whether you re looking to validate your skills or advance your career in the tech industry this certification serves as a critical benchmark for IBM related competencies In today s fast paced tech environment the IBM Solution Developer certification holds significant value for IT professionals and developers aiming to specialize in IBM technologies Designed for individuals who seek a deep understanding of IBM s application development solutions this certification is a gateway to enhancing your professional portfolio As businesses increasingly rely on IBM technologies for

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Technology-Enabled Blended Learning Experiences for Chemistry Education and Outreach Fun Man Fung, Christoph Zimmermann, 2021-07-03 Technology Enabled Blended Learning Experiences for Chemistry Education and Outreach discusses new technologies and their potential for the advancement of chemistry education particularly in topics that are difficult to demonstrate in traditional 2d media The book covers the theoretical background of technologies currently in use such as virtual and augmented reality introducing readers to the current landscape and providing a solid foundation on how technology can be usefully integrated in both learning and teaching chemistry content Other sections cover the implementation of technology how to design a curriculum and how new tactics can be applied to both outreach and evaluation efforts Case studies supplement the information presented providing the reader with practicable examples and applications of covered theories and technologies Drawing on the broad experiences and unique insights of a global team of authors from a whole host of different backgrounds the book aims to stimulate readers creativity and inspire them to find their own novel applications of the techniques highlighted in this volume Provides detailed information on the theoretical background of technology usage in chemistry education including discussions of augmented and virtual reality Helps readers understand available options and make informed decisions on how to best utilize technology to enhance their chemistry teaching using concepts surrounding blended learning Presents examples of theory in practice through case studies that detail completed implementations from around the world Organic Chemistry Michael B. Smith, 2016-03-09 Based on the premise that many if not most reactions in organic chemistry can be explained by variations of fundamental acid base concepts Organic Chemistry An Acid Base Approach provides a framework for understanding the subject that goes beyond mere memorization Using several techniques to develop a relational understanding it helps students fully grasp the essential concepts at the root of organic chemistry This new edition was rewritten largely with the feedback of students in mind and is also based on the author's classroom experiences using the first edition Highlights of the Second Edition Include Reorganized chapters that improve the presentation of material Coverage of new topics such as green chemistry Adding

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closing 26 11 2023 running time 2h40 inc interval based on the novel by paul gallico and adapted for the stage by richard taylor the go between and rachel wagstaff the mirror crack d flowers for mrs harris is a celebration of kindness friendship and

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