

B) Predict the products of the reaction and balance the equation.

C) What is one observation you would expect to make during this reaction?

D) Explain why a reaction occurs.

1. Predict the products of, and write balanced equations for, the following reactions if the reaction occurs. For any reaction that does not occur simply state "no reaction". (14T)

Synthesis Reactions:



Decomposition Reactions:



Single Displacements:



Double Displacements:



1. Predict the products and write a balanced chemical equation for the following chemical reaction: Aqueous aluminum chloride is added to a solution suspected to contain sodium carbonate. (2T)

1. Predict the products and write a balanced chemical equation for the following chemical reaction: Pure methane gas is burned in a barbecue. (2T)

1. Complete the following reactions and balance. (5T)



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S Ashworth



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