

Polyatomic Ion Worksheet

Place the charges on each ion – Write the Formula – Name the Compound

Remember – The cation always goes first.

Example:	NH_4^{+1}	&	I^{-1}	Formula:	<u>NH_4I</u>	Name:	<u>Ammonium Iodide</u>
1.	NH_4	&	O	Formula:	_____	Name:	_____
2.	Ca	&	CO_3	Formula:	_____	Name:	_____
3.	HCO_3	&	Na	Formula:	_____	Name:	_____
4.	NO_3	&	Mg	Formula:	_____	Name:	_____
5.	K	&	SO_4	Formula:	_____	Name:	_____
6.	Li	&	PO_4	Formula:	_____	Name:	_____

Finding the Formula from the Name

Example:	Lithium Chloride	Ions:	<u>Li^{+1} & Cl^{-1}</u>	Formula:	<u>LiCl</u>
1.	Sodium Oxide	Ions:	_____	Formula:	_____
2.	Potassium Bromide	Ions:	_____	Formula:	_____
3.	Calcium Nitride	Ions:	_____	Formula:	_____
4.	Beryllium Sulfide	Ions:	_____	Formula:	_____
5.	Magnesium Iodide	Ions:	_____	Formula:	_____
6.	Aluminum Oxide	Ions:	_____	Formula:	_____
7.	Ammonium Bromide	Ions:	_____	Formula:	_____
8.	Calcium Bicarbonate	Ions:	_____	Formula:	_____
9.	Potassium Phosphate	Ions:	_____	Formula:	_____
10.	Aluminum Sulfate	Ions:	_____	Formula:	_____

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Rachel S Tattersall



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