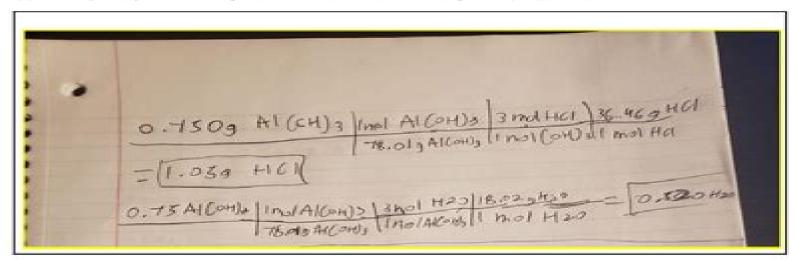
Chemical Equations and Stoichiometry

STOICHIOMETRY PROBLEMS

General Stoichiometry

- Several brands of antacid tablets use aluminum hydroxide to neutralize excess acid.
 Al(OH)₃(s) + 3 HCl(aq) → AlCl₃(aq) + 3 H₂O(1)
 - (a) What quantity of HCl, in grams, can a tablet with 0.750 g of Al(OH); consume?



(b) What quantity, in grams, of water is produced?

•	
	0.750g AI (CH) 3 [mol AI (OH) 3 3 mol HCI 36.46g HCI
	1-16.01 J Alon 2
	-[1.039 HC]
	0.75 AI (OH) 1 Inst AI (OH) > 1 8 MOI H20 18.02 9 H30 - 0.520 H30 18 MOI H20 - 0.520 H30 18 MOI H20

On Your Own Basic Stoichiometry

E Durkheim

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