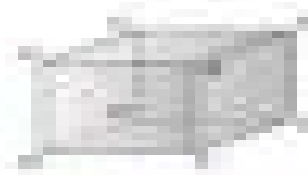


Answer following



- 1. Substrate and temperature:  $20^{\circ}\text{C}$   $30^{\circ}\text{C}$   $40^{\circ}\text{C}$
- 2. A graph of initial velocity:  $100\text{ mg/min}$   $200\text{ mg/min}$   $300\text{ mg/min}$
- 3. A graph of initial velocity:  $100\text{ mg/min}$   $200\text{ mg/min}$   $300\text{ mg/min}$

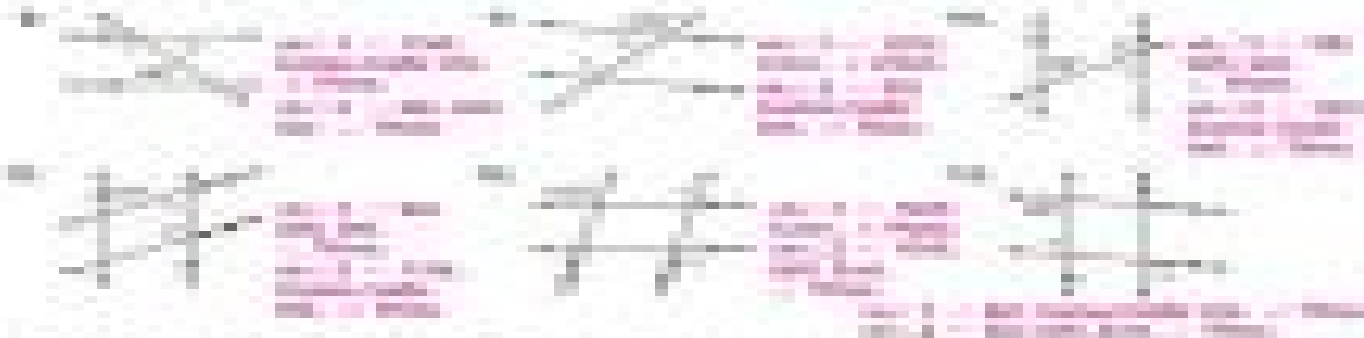
Calculate all values of  $K_m$  and  $V_{max}$  for the following reaction. Assume Michaelis-Menten kinetics and that the reaction is reversible.

- 1.  $K_m = 100\text{ mg/min}$   $V_{max} = 300\text{ mg/min}$
- 2.  $K_m = 100\text{ mg/min}$   $V_{max} = 300\text{ mg/min}$
- 3.  $K_m = 100\text{ mg/min}$   $V_{max} = 300\text{ mg/min}$
- 4.  $K_m = 100\text{ mg/min}$   $V_{max} = 300\text{ mg/min}$



Answer 100

Graph the following data. Assume the Michaelis-Menten model and that the reaction is reversible.



Graph the following data. Assume the Michaelis-Menten model and that the reaction is reversible.



Michaelis-Menten model and that the reaction is reversible.

# Prentice Hall Geometry 10 8 Practice Answers

**Elçi, Atilla, Pieprzyk, Josef, Chefranov,  
Alexander G., Orgun, Mehmet A., Wang,  
Huaxiong, Shankaran, Rajan**

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