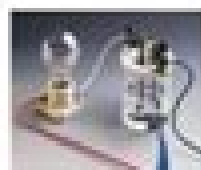
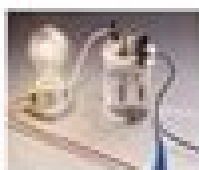


# Chapter Four

## Chemical Reactions in Aqueous Solutions



(a) 1 M  $\text{CH}_3\text{COOH}$   
Nonelectrolyte  
Solution consists  
of molecules  
no ions



(b) 1 M  $\text{NaCl(aq)}$   
Strong electrolyte  
Solution consists of ions



(c) 1 M  $\text{CH}_3\text{COONH}_4\text{(aq)}$   
Weak electrolyte Solution contains  
mostly of molecules, some ions

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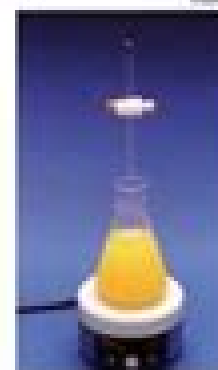
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(a)



(b)



(c)

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# Prentice Hall Chemistry Answer Key Section 4

**Brendan G. Carr**



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