

- d. Secondary passive transport protein
10. An intermediate compound X is formed during the production of urea through *ammonia/carbon dioxide* technology. What is the molecular formula of X?
- $\text{NH}_2\text{COONH}_2$
 - $\text{NH}_2\text{COONH}_4$
 - $\text{NH}_4\text{COONH}_2$
 - $\text{NH}_4\text{COONH}_4$
11. Which of the following are recycled in the manufacture of sodium Carbonate by Solvay's process?
- CO_2 and NH_4Cl
 - CO_2 and NH_3
 - NaCl and CaO
 - NaCl and NH_3

Group 'B'

Give short answer to the following questions.

(8 × 5=40)

- I. An element X has 2 electrons in K shell, 8 electrons in L shell and 5 electrons in M shell.
- Identify the element X and write the number of protons and electrons in it. [3]
 - Size of X^{3-} ion is greater than that of X atom though both contain the same number of protons. Give reason [1]
 - Write down the formula of one of the compounds of X where X is in -3 oxidation state. [1]

OR

Know-how about ionization energy (IE) of elements is crucial aspect in the study of chemical bonding whether they form ionic or covalent bond. The first ionization energies of period second elements are given below

of elements								
ionization energy(kJ/mol)								

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Table of Contents Ongc Model Question Paper For Chemistry

1. Understanding the eBook Ongc Model Question Paper For Chemistry
 - The Rise of Digital Reading Ongc Model Question Paper For Chemistry
 - Advantages of eBooks Over Traditional Books
2. Identifying Ongc Model Question Paper For Chemistry
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ongc Model Question Paper For Chemistry
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ongc Model Question Paper For Chemistry
 - Personalized Recommendations
 - Ongc Model Question Paper For Chemistry User Reviews and Ratings
 - Ongc Model Question Paper For Chemistry and Bestseller Lists

5. Accessing Ongc Model Question Paper For Chemistry Free and Paid eBooks
 - Ongc Model Question Paper For Chemistry Public Domain eBooks
 - Ongc Model Question Paper For Chemistry eBook Subscription Services
 - Ongc Model Question Paper For Chemistry Budget-Friendly Options
6. Navigating Ongc Model Question Paper For Chemistry eBook Formats
 - ePub, PDF, MOBI, and More
 - Ongc Model Question Paper For Chemistry Compatibility with Devices
 - Ongc Model Question Paper For Chemistry Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ongc Model Question Paper For Chemistry
 - Highlighting and Note-Taking Ongc Model Question Paper For Chemistry
 - Interactive Elements Ongc Model Question Paper For Chemistry
8. Staying Engaged with Ongc Model Question Paper For Chemistry
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ongc Model Question Paper For Chemistry
9. Balancing eBooks and Physical Books Ongc Model Question Paper For Chemistry
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ongc Model Question Paper For Chemistry
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Ongc Model Question Paper For Chemistry
 - Setting Reading Goals Ongc Model Question Paper For Chemistry
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Ongc Model Question Paper For Chemistry
 - Fact-Checking eBook Content of Ongc Model Question Paper For Chemistry
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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