

The questions in the test bank cover the concepts from the lessons in Chapter 2. The types of questions include multiple choice, true/false, fill-in-the-blank, and short answer.

Chapter II : Atoms, Molecules, and Ions

- Atomic Structure

- The Iron(II) ion has:
 A) 28 electrons and a charge of 3+ B) 24 electrons and a charge of 3-
 C) 26 electrons and a charge of 2+ D) 24 electrons and a charge of 2+
 E) 26 electrons and a charge of 3+
- Determine the number of *electrons* , *protons* , and *neutrons* of $^{56}_{26}\text{Fe}^{3+}$ and $^{204}_{81}\text{Tl}$?
- A neutral atom of the element potassium $^{39}_{19}\text{K}$ has :
 A. 19 e, 19 p, 21 n B. 40 e, 40 p, 19 n
 C. 19 e, 21 p, 19 n D. 40 e, 19 p, 21 n E. 21 e, 19 p, 19 n
- The number of *neutrons* in an atom of $^{127}_{53}\text{I}$ is :
 A. 53 B. 74 C. 127 D. 180
- An atom of the isotope chlorine-39 consists of how many *protons*, *neutrons*, and *electrons*? (p = proton, n= neutron, e = electron)
 A. 17 p, 17 n, 22 e B. 39 p, 38 n, 39 e
 C. 22 p, 17 n, 17 e D. 17 p, 39 n, 17 e E. 17 p, 22 n, 17 e
- How many **protons** and **electrons** are present in a $^{40}_{20}\text{Ca}^{2+}$ ion?
 A.18 p, 20 e B.20 p, 18 e C.20 p, 21 e D. 20 p, 19 e E. 22 p, 20 e
- The elements in a *row* of the *periodic table* are known as:
 A. metalloids B. noble gases C. a group D. nonmetals E. a period
- True or False :
 () The number of electrons in natural atom equal to the number of neutrons

- Isotopes

- Two *isotopes* of an element *differ* in their:
 A. atomic mass B. atomic number C. number of protons D. number of electrons
- For the following data about unknown element :
 92.21% (27.977) 4.700% (28.976) 3.090% (29.974)

Identify the element ?

Science Focus 2 Test Chapter Atoms

RS Peters



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