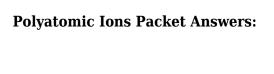
# List of Common Polyatomic Ions

| Cations:<br>Symbol                | Name         | Charge | Symbol                           | Name          | Charge |
|-----------------------------------|--------------|--------|----------------------------------|---------------|--------|
| NH4°                              | Ammonium     | +1     | $Hg_2^{+2}$                      | Dimercury (I) | +2     |
| Anions:                           |              |        |                                  |               |        |
| Symbol                            | <u>Name</u>  | Charge | Symbol                           | Name          | Charge |
| $C_2H_3O_2$                       | Acetate      | - 1    | CO <sub>3</sub> -2               | Carbonate     | -2     |
| HCO <sub>3</sub>                  | Bicarbonate  | - 1    | $CrO_4^{-2}$                     | Chromate      | -2     |
| BrO <sub>3</sub>                  | Bromate      | - 1    | $Cr_2O_7^{-2}$                   | Dichromate    | -2     |
| BrO <sub>2</sub>                  | Bromite      | - 1    | $C_2O_4^{-2}$                    | Oxalate       | -2     |
| ClO <sub>4</sub>                  | Perchlorate  | - 1    | $O_2^{-2}$                       | Peroxide      | -2     |
| ClO <sub>3</sub>                  | Chlorate     | - 1    | SO <sub>4</sub> -2               | Sulfate       | -2     |
| CIO <sub>2</sub>                  | Chlorite     | - 1    | SO <sub>3</sub> -2               | Sulfite       | -2     |
| CIO                               | Hypochlorite | - 1    | S <sub>2</sub> O <sub>3</sub> -2 | Thiosulfate   | -2     |
| CN                                | Cyanide      | - 1    | SiO <sub>3</sub> 2               | Silicate      | -2     |
| SCN                               | Thiocyanate  | - 1    | $MoO_4^{-2}$                     | Molybdate     | -2     |
| OH                                | Hydroxide    | - 1    | $WO_4^{-2}$                      | Tungstate     | -2     |
| NO <sub>3</sub>                   | Nitrate      | - 1    | $BO_3^{-3}$                      | Borate        | -3     |
| NO2                               | Nitrite      | -1     | $PO_4^{-3}$                      | Phosphate     | -3     |
| MnO <sub>4</sub>                  | Permanganate | - 1    | $PO_{3}^{-3}$                    | Phosphite     | -3     |
| HSO.                              | Bisulfate    | - 1    | AsO <sub>4</sub> -3              | Arsenate      | -3     |
| HSO <sub>3</sub>                  | Bisulfite    | - 1    | AsO <sub>3</sub> -3              | Arsenite      | -3     |
| IO <sub>3</sub>                   | Iodate       | - 1    |                                  |               |        |
| IO4                               | Periodate    | - 1    |                                  |               |        |
| C <sub>6</sub> H <sub>5</sub> COO | Benzoate     | -1     |                                  |               |        |

| Symbol                 | Name                    | Charge |
|------------------------|-------------------------|--------|
| $H_{2}PO_{4}^{-1}$     | Dihydrogen phosphate    | - 1    |
| $HPO_4^{-2}$           | Hydrogen phosphate      | -2     |
| Fe(CN) <sub>6</sub> -3 | Hexacyannoferrate (III) | -3     |
| Fe(CN) <sub>6</sub> -4 | Hexacyannoferrate (II)  | -4     |

# **Polyatomic Ions Packet Answers**

**Aharon Loewenschuss, Y. Marcus** 



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