

6th Unit 1 The Nature of Science

1.1 Scientific Knowledge

I. What is Science?

A. Science is the study of the _____ world.

B. Natural science is divided into 3 categories.

1. Biology- the study of _____ things

2. _____ is the study of Earth.

3. Physical science is the study of _____ things and energy and also includes _____.

4. These categories sometimes _____.

II. What Does Science Tell Us?

A. What you read in your science book is not always _____ by everyone.

B. The _____ of science are simply the most widely accepted explanations.

C. What we learn in science are what most scientists _____ are the best explanations for what happens. These are referred to as _____.

D. A scientific theory is an _____ supported by a large amount of _____.

1. _____ theory states that living things are made up of cells that perform the basic functions of life.

2. _____ tectonics theory states that Earth's surface is made up of plates that _____.

3. Wave theory of _____ states that each color of visible _____ has a wave of a specific _____.

III. How Do Scientific Theories Differ from Laws?

A. Laws describe principles of nature

1. A description of a specific relationship under given conditions in the natural world is referred to as a scientific _____.

2. Scientific laws _____ how the world works. They hold true for anywhere in the _____.

3. _____ law is one example of a scientific law. Boyle's law states that as _____ on a gas increases, it's _____ decreases.

Science Fusion 6th Grade Unit Test

G Thomas



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 $4\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
 $5\text{Pb} + 2\text{HCl} \rightarrow 2\text{H}_2 + \text{PbCl}_2$
 $2\text{Al} + 3\text{K}_2\text{SO}_4 \rightarrow 4\text{K} + 3\text{Al}_2\text{O}_3 + 6\text{S}$
 $4\text{CH}_4 + 7\text{O}_2 \rightarrow 2\text{CO}_2 + 2\text{H}_2\text{O} + 8\text{H}_2\text{O}$
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 $3\text{HBr} + 1\text{KHCO}_3 + 1\text{H}_2\text{O} + 1\text{KBr} + \text{CO}_2 + 2\text{GaBr}_3 + 3\text{Na}_2\text{SO}_3 + 1\text{Ga}_2\text{SO}_3 + 3\text{NaBr} + 3\text{SnO}_2 + 3\text{NF}_3 + 3\text{SnF}_2 + 1\text{N}_2\text{O}_3$ using the following equation $2\text{NaOH} + \text{H}_2\text{SO}_4 \rightarrow 2\text{H}_2\text{O} + \text{Na}_2\text{SO}_4$

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