National Co.	Theter:	
	L. Carlotte	

# 9-3 Review and Reinforcement

### Classifying chemical Reactions

On the line as the ligh, wide the letter of the type of chamical reaction represented by each numerion below.

 $1. A + B \rightarrow AB$ 

a, decomposition reaction.

 $_{---}$  2. A + BX  $\rightarrow$  AX + B

4. All + A + B

b. direct combination reaction

- $3. AX + BY \Rightarrow AY + BX$
- double-replacement reaction
   d. single-replacement reaction

Describe in wards each of the following types of reactions.

- 5. direct combination reaction
- 6. decomposition reaction
- T. stagle-replacement reaction.
- N. double-replacement reaction

Decide whether each of the following equations represents a direct combination (dc), decomposition (d), single-replacement (xr), or double-replacement (dr) reaction. Write year univers on the line.

- $\underline{\hspace{1cm}}$  0.  $CO_2 \rightarrow C + O_2$
- \_\_\_\_\_ ID. NaCl + AgNO<sub>2</sub> → NaNO<sub>2</sub> + AgCl
- $11.5_0 + 401 \rightarrow 8800_0$
- \_\_\_\_\_ 12. BaCl<sub>2</sub> + 2NaOH ⇒ 2NaCl + Ba(OH)<sub>2</sub>
- \_\_\_\_\_ 13. Zn + CuSO<sub>1</sub> → ZnSO<sub>1</sub> + Cu
- \_\_\_\_\_ 14, CH<sub>0</sub> + C + 2H<sub>1</sub>
- \_\_\_\_\_ 15. PbrNO<sub>1</sub>u + Mg → Pb + MgtNO<sub>1</sub>b

Charmistry.

# **Review And Reinforcement Chemical Kinetics Answers**

**David Baud** 

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