

QUESTION 3



GAUTENG PROVINCE
EDUCATION

PROVINCIAL EXAMINATION NOVEMBER 2022 GRADE 11

PHYSICAL SCIENCES: CHEMISTRY
(PAPER 2)

TIME: 2 hours

MARKS: 100

QUESTION 3 (Start on a new page.)

In the table below the boiling points of four hydrogen halides are given.

Hydrogen halides	A	B	C	D
Boiling points (°C)	18,4	-85	-87	-102,3

- 3.1 Define the term boiling point. (1)
- 3.2 What is the relationship between strength of the intermolecular forces and boiling point? (1)
- 3.3 Which of the hydrogen halides (A, B, C or D) in the table above has the ...
 - 3.3.1 highest vapour pressure? (Refer to the boiling point in the table above to give a reason for the answer.) (2)
 - 3.3.2 strongest intermolecular forces? (1)
- 3.4 Draw a line graph to represent the boiling points of the hydrogen halides. Label the axes clearly and give the graph a heading. (5)
- 3.5 Compare compounds A and B. Refer to the TYPE and the STRENGTH of intermolecular forces and ENERGY needed to explain the difference in boiling points between hydrogen halides. (4)

(14)

PHYSICAL PROPERTIES INTERMOLECULAR FORCES

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