

Periodic Table of the Elements

Periodic Table of the Elements																														
	1	2											3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18		
1	H 1.008	He 4.003											Li 6.941	Be 9.012	B 10.81	C 12.01	N 14.01	O 16.00	F 18.99	Ne 20.18										
2	Na 22.99	Mg 24.31											Al 26.98	Si 28.09	P 30.97	S 32.07	Cl 35.45	Ar 39.95												
3	K 39.10	Ca 40.08	Sc 44.96	Ti 47.88	V 50.94	Cr 52.00	Mn 54.94	Fe 55.85	Co 58.93	Ni 58.71	Cu 63.55	Zn 65.38	Ga 69.72	Ge 72.64	As 74.92	Se 78.96	Br 79.90	Kr 83.80												
4	Rb 85.47	Sr 87.62	Y 88.91	Zr 91.22	Nb 92.91	Mo 95.94	Tc 98.91	Ru 101.07	Rh 102.91	Pd 106.42	Ag 107.87	Cd 112.41	In 114.82	Sn 118.71	Sb 121.76	Te 127.60	I 126.91	Xe 131.29												
5	Cs 132.91	Ba 137.33	La 138.91	Ce 140.12	Pr 140.91	Nd 144.24	Pm 144.91	Sm 150.36	Eu 151.96	Gd 157.25	Tb 158.93	Dy 162.50	Ho 164.93	Er 167.26	Tm 168.93	Yb 173.05	Lu 174.97													
6	Fr [223]	Ra [226]	Ac [227]	Th [232]	Pa [231]	U [238]	Np [237]	Pu [244]	Am [243]	Cm [247]	Bk [247]	Cf [251]	Es [252]	Fm [257]	Mn [288]	Ni [289]	Pu [289]	Am [287]	Cm [284]	Bk [284]	Cf [281]	Es [277]	Fm [277]	Mn [270]	Ni [271]	Pu [271]	Am [269]	Cm [267]	Bk [264]	Cf [264]
7	Fr [223]	Ra [226]	Ac [227]	Th [232]	Pa [231]	U [238]	Np [237]	Pu [244]	Am [243]	Cm [247]	Bk [247]	Cf [251]	Es [252]	Fm [257]	Mn [288]	Ni [289]	Pu [289]	Am [287]	Cm [284]	Bk [284]	Cf [281]	Es [277]	Fm [277]	Mn [270]	Ni [271]	Pu [271]	Am [269]	Cm [267]	Bk [264]	Cf [264]
8	Ce [140.12]	Pr [140.91]	Nd [144.24]	Pm [144.91]	Sm [150.36]	Eu [151.96]	Gd [157.25]	Tb [158.93]	Dy [162.50]	Ho [164.93]	Er [167.26]	Tm [168.93]	Yb [173.05]	Lu [174.97]																
	Th [232]	Pa [231]	Np [237]	Pu [244]	Am [243]	Cm [247]	Bk [247]	Cf [251]	Es [252]	Fm [257]	Mn [288]	Ni [289]	Pu [289]	Am [287]	Cm [284]	Bk [284]	Cf [281]	Es [277]	Fm [277]	Mn [270]	Ni [271]	Pu [271]	Am [269]	Cm [267]	Bk [264]	Cf [264]				

The only metal that is liquid at room temperature.

Period 3
element with
7 valence
electrons.

Semi-conductor
metalloid used
in electronics.

Only one valence electron, but not part of a family.

Extremely reactive non-metal.

Group that will form 2+ cations.

The only non-metal that is liquid at room temperature.

Completely
un-reactive family
with complete
valence shell.

Extremely reactive metal.

Group with 4 valence electrons.

Periodic Trends Review

Liyong Dong



Periodic Trends Review:

Periodic Trends Review Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Periodic Trends Review**," published by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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