## Periodic Table Named-Groups Worksheet – 2

**DIRECTIONS:** Use the Periodic Table shown below to answer each question, <u>using the letters in</u> each "box" to identify the required element(s).

A		7							_			
		1							D		E	
C												
	E				G			H				P
	L									I		Г
J		Z		M			N				0	F
o l		\B\		s	т	0	id					
		\.	\	100		20		50 58			110 19	
		1	v				w					
		- 1	×									

#### List elements in alphabetical order when more than 1 element answers a question.

	ment A is in a box all by itself because: A) it is really a Noble Gas		
gas	<ul> <li>b) it does NOT share the properties of the other elements in its group</li> </ul>	E) it is a liq	uid

 <ol><li>Which of these elements is/are</li></ol>	highly reactive?	A) C	B) "E"	C) . K.	D) A, B, and C	E) A and B only

3) What is the letter of the element whose atomic number is <u>closest</u> to the element labeled "z"?

4) Which letter labels the halogen element with the lowest atomic number?

5) Which of these elements is the least reactive? A) "o" B) "x" C) "A" D) "o" E) "B"

6) Which of these elements is/are <u>NOT</u> transition elements/metals (list ALL that are not transition elements)? A) "6" B) "N" C) "N" D) "N" E) "2"

continued on other side

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**D** Kirk

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