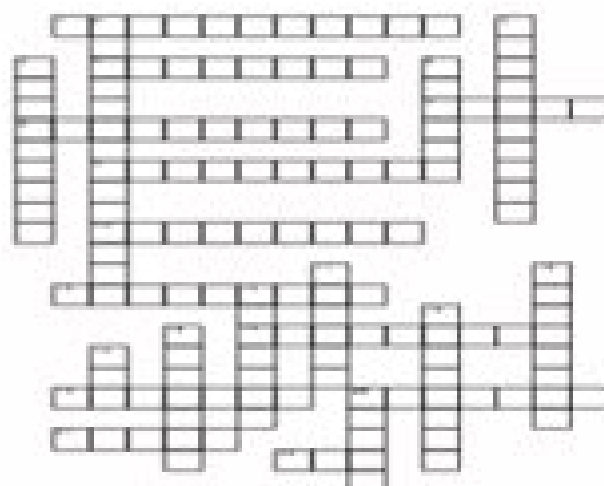


## Periodic Table Assignment

Read pages 14 – 19. Complete the crossword puzzle (in pencil) and fill in the period table as instructed below.



1000

1. Synthesis elements with atomic numbers greater than 93
9. This law states that the properties of elements repeat as a function of their atomic number
7. Lavoisier is in this period
8. Elements were originally placed in order of atomic \_\_\_\_, now they are in order of atomic \_\_\_\_
6. The group that consists of very unreactive gases
10. Number of elements known in 1869
13. Element 101 was named in his honour
17. The periodic table is normally drawn with 18 columns. A more concise representation has \_\_\_\_ columns
19. Most nonmetals are in this state
20. Nonmetal that is a liquid at SATP
21. The general name given to a column in the periodic table
22. According to figure 6, when aluminum combines with oxygen, three oxygen atoms combine with this many aluminum atoms.

1999

2. Elements that most closely follow the periodic law
3. The first scientist to organize elements based on repeating patterns
4. The modern name for alkali
6. Most elements are \_\_\_\_\_ since they fall to the left of the "staircase line"
11. The general name given to a row (left-to-right) in the periodic table
12. Iodine belongs to this group
14. The first alkali metal
15. This element was assigned a relative atomic mass of one because it is the lightest element
16. The state of matter represented by (aq)
18. Silicon is in this group
20. The staircase that separates metals and nonmetals begins below this element

1	0.0	0																	0.0	0		
2	0.0	0	0.0	0																	0.0	0
3	0.0	0	0.0	0																	0.0	0
4	0.0	0	0.0	0																	0.0	0
5	0.0	0	0.0	0																	0.0	0
6	0.0	0	0.0	0																	0.0	0
7	0.0	0	0.0	0																	0.0	0

- 1) Place a large G in boxes that house elements that are gases at S.A.T.P., and a large L in boxes that contain elements that are liquids. The rest of the elements are solids; do not write anything in these boxes.
- 2) Darken the "staircase line" that divides metals from non-metals.
- 3) Metalloids (i.e. semi-conductors) are elements that have properties midway between metals and non-metals. The metalloids are: B, Si, Ge, As, Sb, Te, Po, At. Identify the metalloids by drawing diagonal lines in these boxes.
- 4) Notice that each group (column) in the periodic table is identified at the top with a number and/or letter. There are two conventions used in the textbook. Label the groups that end in A (e.g. 1A, 2A, 3A, etc.).
- 5) Frame and colour the following groups (note: H is not an 1A metal): Alkali metals, alkaline earth metals, halogens, noble gases, transition metals, the inner transition elements (made up of lanthanides and actinides).
- 6) Create a legend for your periodic table.

# Periodic Table Assignment Answer Key

**TJ McKenna**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the name TJ McKenna.

## **Periodic Table Assignment Answer Key:**

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**College Chemistry I** Norman H. Nachtrieb, 1960      **Learning Chemistry 7 Solution Book (Year 2023-24)**, 2024-01-02      **Learning Elementary Chemistry Class 7 Teacher Resource Book (Academic Year 2023-24)**, 2023-05-20 Learning Elementary Chemistry Class 7 Teacher Resource Book Academic Year 2023 24      [Chemistry in the Community \(ChemCom\)](#) American Chemical Society, 2011-06-17 Touted as the most successful NSF funded project published Chemistry in the Community ChemCom by the American Chemical Society ACS offers a meaningful and memorable chemistry program for all levels of high school students ChemCom covers traditional chemistry topics within the context of societal issues and real world scenarios Centered on decision making activities where students are responsible for generating data in an investigating analyzing that data and then applying their chemistry knowledge to solve the presented problem The text is intensively laboratory based with all 39 of the investigations integrated within the text not separate from the reading With the ChemCom program students learn more organic and biochemistry more environmental and industrial chemistry and more on the particulate nature of matter than other textbooks all within the relevance of solving problems that arise in everyday life Meticulously updated to meet the needs of today s teachers and students the new sixth edition of ChemCom adheres to the new science framework as well as the forthcoming next generation of science standards Incorporating advances in learning and cognitive sciences ChemCom s wide ranging coverage builds upon the concepts and principles found in the National Science Education Standards Correlations are available showing how closely aligned ChemCom is to these and other state standards      **Holt Chemistry** Ralph Thomas Myers, 2004      **Formative Assessment in a Brain-Compatible Classroom** Marcia L. Tate, 2024-05-21 Students learn in diverse ways and their grades may not always accurately reflect their true understanding Marcia L Tate explores the what why and how of assessment to help educators know whether their students are truly learning Dive into Tate s insightful exploration of building student confidence and nurturing organic student retention and comprehension Grades K 12 teachers will Learn how to create a brain compatible classroom that fosters success and confidence in students when they are assessed Plan lessons so students know what they should know and be able to do Ask effective questions to find out if students are really learning what they

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College Chemistry II Norman H. Nachtrieb, 1960 **Inorganic Chemistry-II (For M.Sc. Course for Universities in Uttarakhand)** Dr. Geeta Tiwari, This book entitled Inorganic Chemistry II is an effort to present the subject matter in a comprehensible and easily understandable form This textbook is purposefully prepared for the postgraduate Inorganic Chemistry second semester course and it covers all the topics recommended **Science Scope**, 2003 **THE Journal**, 1999 *Chemistry in the Laboratory* James M. Postma, Julian L. Robert, J. Leland Hollenberg, 2004-03-12 This clearly written class tested manual has long given students hands on experience covering all the essential topics in general chemistry Stand alone experiments provide all the background introduction necessary to work with any general chemistry text This revised edition offers new experiments and expanded information on applications to real world situations Study Guide Woolfolk, 2000-06 **Biology/science Materials** Carolina Biological Supply Company, 1991 **Making Sense of Sensemaking** TJ McKenna, 2025 Dive into the transformative world of science education with this groundbreaking guide Learn how to navigate the journey from traditional teaching to a dynamic student centered approach that emphasizes understanding over rote learning Grounded in the latest educational research and aligned with the Next Generation Science Standards NGSS this book provides practical strategies for creating K 12 classrooms where students actively engage in

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*College Reading and Study Skills* Kathleen T. McWhorter, 1983

Armchair Chemistry David Bradley, Joel Levy, 2018-02-13 Part of the Armchair series Armchair Chemistry is a quick refresher course in how we survey of the science It explains how we evolved from believing in alchemy to discovering modern chemical equations and goes into detail about the lives of the scientists that uncovered them Fascinating and interactive this is ideal for the student brushing up on a subject or for as a clear and accessible companion for beginner s and experts alike It contains explanations of different chemical concepts as well as profiles of key scientists and and their discoveries It contains clear and concise explanations of different chemical concepts as well as profiles of key scientists and their discoveries A unique feature of the book is its simple step by step exercises Some of these have everyday applications others are theoretical puzzles but all are designed to challenge you and test your newly acquired knowledge The perfect companion for beginners and experts alike Armchair Chemistry does not assume prior knowledge of the subject It conveys the basic elements of chemistry in a way that is clear and accessible no matter your level of ability

**Creating the High Schools of Our Choice** Tim Westerberg, 2013-09-13 This book provides strategies tools and examples to help high school principals achieve the goals put forward in the No Child Left Behind legislation With its clear examples of best practices in leadership personalization and strategic use of data it will show you how to achieve your school s academic mission

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