

Chapter 1: Limiting Reagents and Percent Yield (Pages 1-3)

Remember: Limiting reagent is the reactant that will completely react and leave another reactant

Excess: All reactants except the limiting reagent will have some left over. Remember: Limiting reagent always determines the amount of product.

Remember: Limiting reagent will determine how much product is produced and

byproducts will be. The limiting reagent is also a reactant.



The ratio required for the composition for the limiting reagent.

2H ₂ (g)	20.00 grams
O ₂ (g)	32.00 grams
2H ₂ O(l)	36.00 grams
H ₂ O	36.00 grams

1. 20.00 g of each element will be used & react to form 36.00 g of product water. 20.00 g of hydrogen & 32.00 g of oxygen will be used. 36.00 g of water is produced.



2. 20.00 g H₂(g) is the limiting reagent. 36.00 g H₂O(l) is the maximum amount.

3. 32.00 g of oxygen is the excess reagent. 36.00 g of water is produced.



4. 36.00 g of water is produced from 20.00 g of hydrogen & 32.00 g of oxygen. 36.00 g of water is the maximum amount. 36.00 g of water is the actual yield. 100% yield.



Percent Yield And Limiting Reactants Quiz

Rachel S Tattersall



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