

Name :

Percent Composition and Empirical Formula

1. What is the percent composition of all the elements in $\text{Al}_2(\text{SO}_4)_3$?

2. A compound contains 36.5% Na, 25.4% S, and 38.1% O. Find its empirical formula.

3. Find the empirical formula of a compound consisting of 53.7% iron and 46.3% sulfur.

4. Determine the empirical formula for the following compounds:
 - a. A compound containing 0.0130 mol carbon, 0.0390 mol hydrogen, and 0.0065 mol oxygen.

 - b. A compound containing 72.7% magnesium and 27.8% nitrogen by mass.

 - c. Glucose contains 40% carbon, 6.7% hydrogen, and 53.3% oxygen by mass.

Percent Composition And Empirical Formulas Answer Key

Rosina Ehmann



Percent Composition And Empirical Formulas Answer Key:

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