

PHYSICAL SCIENCES P.2

MEMO-2024



basic education

Department:
Basic Education
REPUBLIC OF SOUTH AFRICA

G.12

**NATIONAL
SENIOR CERTIFICATE**

Diagram on page 9 is redrawn below for easy reference.



Changes to the conditions of equilibrium are made at different times. The graph shows the results obtained. X, Y and Z represent the gases in the above reaction.



2.1.1 Define for both chemical equilibrium

Use the graph to answer the questions below.

The inorganic reagents shown below are available for reactions IV and Y.

Bz	$\text{H}_2\text{SO}_4(\text{conc})$	$\text{NaOH}(\text{conc})$	HBr	H_2
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4.5 Write down the TYPE of elimination in reaction III.

4.6 Compounds R and T are positional isomers.

4.7 Identify the reagent used in reaction IV. (1)



4.8 Draw the half-cell reaction for the zinc electrode. (1)



4.9 Identify the reaction type.

Write a balanced chemical equation for the reaction.

4.10 Draw a half-cell reaction for the copper electrode. (1)

4.11 The simplified diagram below represents an electrochemical cell. The electrodes are made of copper.

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Scott C. Dulebohn



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 - The Rise of Digital Reading Physical Science Paper 2 2014 September 2014grade 12 Memo
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Paper 2 2014 September 2014grade 12 Memo
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
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 - Popular eBook Platforms
 - Features to Look for in an Physical Science Paper 2 2014 September 2014grade 12 Memo
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Paper 2 2014 September 2014grade 12 Memo
 - Personalized Recommendations
 - Physical Science Paper 2 2014 September 2014grade 12 Memo User Reviews and Ratings
 - Physical Science Paper 2 2014 September 2014grade 12 Memo and Bestseller Lists
5. Accessing Physical Science Paper 2 2014 September 2014grade 12 Memo Free and Paid eBooks
 - Physical Science Paper 2 2014 September 2014grade 12 Memo Public Domain eBooks
 - Physical Science Paper 2 2014 September 2014grade 12 Memo eBook Subscription Services
 - Physical Science Paper 2 2014 September 2014grade 12 Memo Budget-Friendly Options

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 - ePub, PDF, MOBI, and More
 - Physical Science Paper 2 2014 September 2014grade 12 Memo Compatibility with Devices
 - Physical Science Paper 2 2014 September 2014grade 12 Memo Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Science Paper 2 2014 September 2014grade 12 Memo
 - Highlighting and Note-Taking Physical Science Paper 2 2014 September 2014grade 12 Memo
 - Interactive Elements Physical Science Paper 2 2014 September 2014grade 12 Memo
8. Staying Engaged with Physical Science Paper 2 2014 September 2014grade 12 Memo
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science Paper 2 2014 September 2014grade 12 Memo
9. Balancing eBooks and Physical Books Physical Science Paper 2 2014 September 2014grade 12 Memo
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science Paper 2 2014 September 2014grade 12 Memo
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Physical Science Paper 2 2014 September 2014grade 12 Memo
 - Setting Reading Goals Physical Science Paper 2 2014 September 2014grade 12 Memo
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Physical Science Paper 2 2014 September 2014grade 12 Memo
 - Fact-Checking eBook Content of Physical Science Paper 2 2014 September 2014grade 12 Memo
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
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