

# BALANCING CHEMICAL EQUATIONS

Name Grove

Rewrite and balance the equations below.

1.  $\text{N}_2 + \text{H}_2 \rightarrow \text{NH}_3$   $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$
2.  $\text{KClO}_3 \rightarrow \text{KCl} + \text{O}_2$   $2\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2$
3.  $\text{NaCl} + \text{F}_2 \rightarrow \text{NaF} + \text{Cl}_2$   $2\text{NaCl} + \text{F}_2 \rightarrow 2\text{NaF} + \text{Cl}_2$
4.  $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$   $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
5.  $\text{AgNO}_3 + \text{MgCl}_2 \rightarrow \text{AgCl} + \text{Mg}(\text{NO}_3)_2$   $2\text{AgNO}_3 + \text{MgCl}_2 \rightarrow 2\text{AgCl} + \text{Mg}(\text{NO}_3)_2$
6.  $\text{AlBr}_3 + \text{K}_2\text{SO}_4 \rightarrow \text{KBr} + \text{Al}_2(\text{SO}_4)_3$   $2\text{AlBr}_3 + 3\text{K}_2\text{SO}_4 \rightarrow 6\text{KBr} + \text{Al}_2(\text{SO}_4)_3$
7.  $\text{CH}_4 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$   $\text{CH}_4 + 2\text{O}_2 \rightarrow \text{CO}_2 + 2\text{H}_2\text{O}$
8.  $\text{C}_3\text{H}_8 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$   $\text{C}_3\text{H}_8 + 5\text{O}_2 \rightarrow 3\text{CO}_2 + 4\text{H}_2\text{O}$
9.  $\text{C}_8\text{H}_{18} + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$   $\text{C}_8\text{H}_{18} + \frac{25}{2}\text{O}_2 \rightarrow 8\text{CO}_2 + 9\text{H}_2\text{O}$
10.  $\text{FeCl}_3 + \text{NaOH} \rightarrow \text{Fe}(\text{OH})_3 + \text{NaCl}$   $\text{FeCl}_3 + 3\text{NaOH} \rightarrow \text{Fe}(\text{OH})_3 + 3\text{NaCl}$
11.  $\text{P} + \text{O}_2 \rightarrow \text{P}_2\text{O}_5$   $4\text{P} + 5\text{O}_2 \rightarrow 2\text{P}_2\text{O}_5$
12.  $\text{Na} + \text{H}_2\text{O} \rightarrow \text{NaOH} + \text{H}_2$   $2\text{Na} + 2\text{H}_2\text{O} \rightarrow 2\text{NaOH} + \text{H}_2$
13.  $\text{Ag}_2\text{O} \rightarrow \text{Ag} + \text{O}_2$   $2\text{Ag}_2\text{O} \rightarrow 4\text{Ag} + \text{O}_2$
14.  $\text{S}_8 + \text{O}_2 \rightarrow \text{SO}_3$   $\text{S}_8 + 12\text{O}_2 \rightarrow 8\text{SO}_3$
15.  $\text{CO}_2 + \text{H}_2\text{O} \rightarrow \text{C}_6\text{H}_{12}\text{O}_6 + \text{O}_2$   $6\text{CO}_2 + 6\text{H}_2\text{O} \rightarrow \text{C}_6\text{H}_{12}\text{O}_6 + 6\text{O}_2$
16.  $\text{K} + \text{MgBr}_2 \rightarrow \text{KBr} + \text{Mg}$   $2\text{K} + \text{MgBr}_2 \rightarrow 2\text{KBr} + \text{Mg}$
17.  $\text{HCl} + \text{CaCO}_3 \rightarrow \text{CaCl}_2 + \text{H}_2\text{O} + \text{CO}_2$   $2\text{HCl} + \text{CaCO}_3 \rightarrow \text{CaCl}_2 + \text{H}_2\text{O} + \text{CO}_2$

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**David Kirk**



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## **Table of Contents Physical Science If8767 Answers Balancing Chemical**

1. Understanding the eBook Physical Science If8767 Answers Balancing Chemical
  - The Rise of Digital Reading Physical Science If8767 Answers Balancing Chemical
  - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science If8767 Answers Balancing Chemical
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Physical Science If8767 Answers Balancing Chemical
  - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science If8767 Answers Balancing Chemical
  - Personalized Recommendations
  - Physical Science If8767 Answers Balancing Chemical User Reviews and Ratings
  - Physical Science If8767 Answers Balancing Chemical and Bestseller Lists
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  - Physical Science If8767 Answers Balancing Chemical Budget-Friendly Options
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    - Physical Science If8767 Answers Balancing Chemical Enhanced eBook Features
  7. Enhancing Your Reading Experience
    - Adjustable Fonts and Text Sizes of Physical Science If8767 Answers Balancing Chemical
    - Highlighting and Note-Taking Physical Science If8767 Answers Balancing Chemical
    - Interactive Elements Physical Science If8767 Answers Balancing Chemical
  8. Staying Engaged with Physical Science If8767 Answers Balancing Chemical
    - Joining Online Reading Communities
    - Participating in Virtual Book Clubs
    - Following Authors and Publishers Physical Science If8767 Answers Balancing Chemical
  9. Balancing eBooks and Physical Books Physical Science If8767 Answers Balancing Chemical
    - Benefits of a Digital Library
    - Creating a Diverse Reading Collection Physical Science If8767 Answers Balancing Chemical
  10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
  11. Cultivating a Reading Routine Physical Science If8767 Answers Balancing Chemical
    - Setting Reading Goals Physical Science If8767 Answers Balancing Chemical
    - Carving Out Dedicated Reading Time
  12. Sourcing Reliable Information of Physical Science If8767 Answers Balancing Chemical
    - Fact-Checking eBook Content of Physical Science If8767 Answers Balancing Chemical
    - Distinguishing Credible Sources
  13. Promoting Lifelong Learning
    - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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