

# Bonding and Reactions

## Study Guide

### 10 pages including KEY

#### Chemical Bonding and Reactions Study Guide

This study packet will help you study for the Chemical Bonding and Reactions exam by highlighting the most important concepts on the exam. This packet highlights key concepts and their correct answer on the exam and should be used as a guide to the material you should know.

#### B. VOCABULARY

Atom	Electron	Electron transfer	Electron transfer
Atomic mass	Electron	Electron transfer	Electron transfer
Atomic weight	Electron	Electron transfer	Electron transfer
Atomic	Electron	Electron transfer	Electron transfer
Chemical bond	Electron	Electron transfer	Electron transfer
Chemical reaction	Electron	Electron transfer	Electron transfer

#### C. READING

1. Why do atoms bond together?
2. What is a covalent bond?
3. What kind of electron flow is covalent bond?
4. What is an ionic bond?
5. What kind of electron flow is ionic bond?
6. Why don't noble gases bond together with other elements?

Now do the following for the following compounds with multiple valence electrons:

1. Oxygen (O<sub>2</sub>)
2. Oxygen (O<sub>2</sub>)

Now do the following compounds with multiple valence electrons:

1. Water (H<sub>2</sub>O)
2. Ammonia (NH<sub>3</sub>)

#### REACTIONS

What happens in a chemical reaction?

What does a chemical reaction look like?

What is a chemical reaction?

What is a chemical reaction?

What is the law of conservation of mass?

What is the law of conservation of mass?

A chemical reaction is a process in which a chemical bond is broken and a chemical reaction is formed. The chemical reaction is a process in which a chemical bond is broken and a chemical reaction is formed. The chemical reaction is a process in which a chemical bond is broken and a chemical reaction is formed.

What is a chemical reaction? A chemical reaction is a process in which a chemical bond is broken and a chemical reaction is formed. The chemical reaction is a process in which a chemical bond is broken and a chemical reaction is formed.

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#### B. VOCABULARY

1. Atom
2. Atomic mass
3. Atomic weight
4. Chemical bond
5. Chemical reaction
6. Electron
7. Electron transfer
8. Electron transfer
9. Electron transfer
10. Electron transfer
11. Electron transfer
12. Electron transfer
13. Electron transfer
14. Electron transfer
15. Electron transfer
16. Electron transfer
17. Electron transfer
18. Electron transfer
19. Electron transfer
20. Electron transfer

#### Bonding and Chemical Reactions - KEY

#### C. READING

1. Why do atoms bond together?
  - Atoms bond together to achieve a more stable electronic configuration, typically by filling or emptying their valence electron shells to satisfy the octet rule.
2. What is a covalent bond?
  - A covalent bond is a type of chemical bond formed between two atoms by the sharing of electron pairs. It is typically formed between non-metal atoms.
3. What kind of electron flow is covalent bond?
  - In a covalent bond, the electron pairs are shared between the two atoms, resulting in a mutual attraction that holds the atoms together.
4. What is an ionic bond?
  - An ionic bond is a type of chemical bond formed between a metal atom and a non-metal atom. It is formed by the transfer of electrons from the metal atom to the non-metal atom, resulting in the formation of oppositely charged ions that are held together by electrostatic attraction.
5. What kind of electron flow is ionic bond?
  - In an ionic bond, the electron pairs are transferred from the metal atom to the non-metal atom, resulting in the formation of oppositely charged ions that are held together by electrostatic attraction.
6. Why don't noble gases bond together with other elements?
  - Noble gases have a full valence shell, which makes them chemically stable and unreactive with other elements.
7. What is a chemical reaction?
  - A chemical reaction is a process in which a chemical bond is broken and a chemical reaction is formed. The chemical reaction is a process in which a chemical bond is broken and a chemical reaction is formed.
8. What is a chemical reaction?
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20. What is a chemical reaction?
  - A chemical reaction is a process in which a chemical bond is broken and a chemical reaction is formed. The chemical reaction is a process in which a chemical bond is broken and a chemical reaction is formed.

# Physical Science Bonding And Chemical Reactions Test

**Craig T. Smith**



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