

T.Y.B.Sc. Semester 5 Mock Exam Question Paper
Chemistry Paper I – Physical Chemistry
Multiple Choice Questions

- Q1 Vapour pressure of a liquid solventwhen a non volatile solute is dissolve in it. 1M
- a) Increase
 - b) Decrease
 - c) no change
 - d) Expand
- Q2 Raoult's law is expressed by.....
- a) $\frac{P^o - P}{P^o} = X_2$
 - b) $\frac{P - P^o}{P^o} = X_2$
 - c) $\frac{P - P^o}{P} = X_2$
 - d) $\frac{P - P^o}{P^o} = X$
- Q3 is used in Rast method because it has a high value of K_f .
- a) Camphor
 - b) Bi phenyl
 - c) Naphthalene
 - d) Sodium chloride
- Q4 Boiling point elevation of the solution ΔT_b is given as.....
- a) $\Delta T_b = T - T_o$
 - b) $\Delta T_b = T_o - T$
 - c) $\Delta T_b = T + T_o$
 - d) $\Delta T_b = T \times T_o$
- Q5 The unit for activation of energy is
- a) kJ mol^{-1}
 - b) kJ K^{-1}
 - c) $\text{kJ K}^{-1} \text{mol}^{-1}$
 - d) kJ R^{-1}

Physical Chemistry Multiple Choice Questions

Dr Kusum Baser



Physical Chemistry Multiple Choice Questions:

Multiple Choice Questions in Physical Chemistry for Undergraduate Students Dr Kusum Baser, 2022-08-25

Objective questions are very much useful for practice of particular topic Multiple choice questions are important to practice any topic to solve objective questions very deep study and knowledge of specific topic is needed This book helps the students to do practice in such a way so they can do preparation for competitive exams This book covers important topics of physical chemistry for undergraduate students

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Working Method Approach for Introductory Physical Chemistry Calculations Brian J Hathaway, Clair Murphy, Brian Murphy, 2019-05-16 A Working Method Approach for Introductory Physical Chemistry Calculations is a concise inexpensive introduction to first year chemistry that is aimed at students who are weak in chemistry or have no chemistry on entry to university Such students usually find physical chemistry the most difficult part of the chemistry course and within this section numerical problem solving is an additional difficulty The text should also be invaluable to first year intending chemists This text provides an introduction to physical chemistry and the gas laws followed by chapters on thermodynamics chemical equilibrium electrochemistry and chemical kinetics Each section involves a brief introduction followed by a representative examination question which is broken down into a proposed working method Both short multiple choice questions and related full examination type questions are included This book will prove invaluable to students who need encouragement in a logical

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